



Computer Simulations of Biomolecular Systems

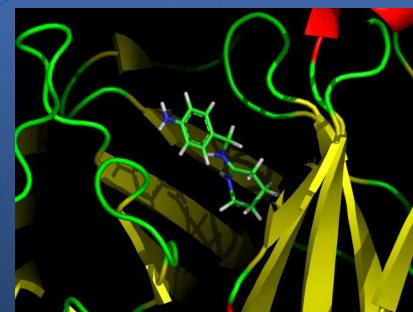
Jonathan W Essex

University of Southampton



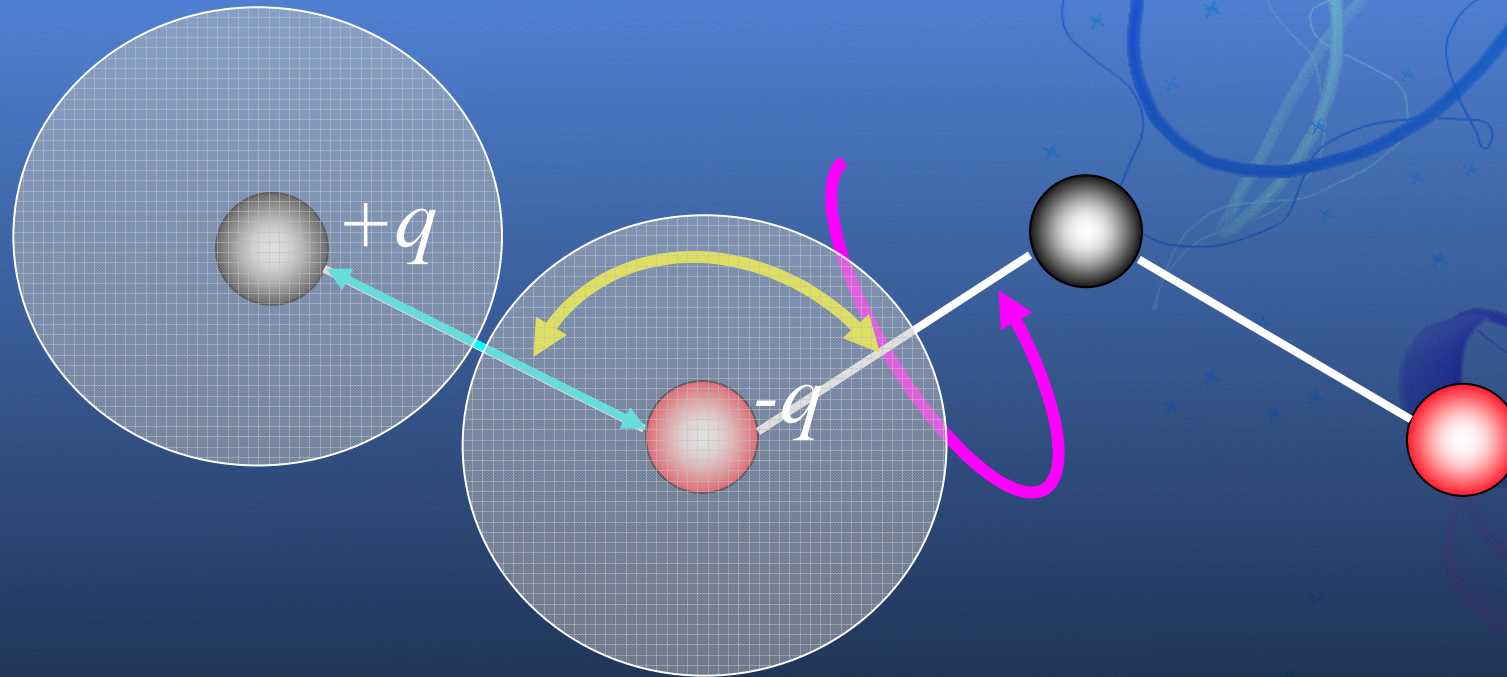
Introduction

- Structure-Based Virtual Screening (SVS) is a widely-used technique for lead discovery and optimisation
 - Protein-Ligand Docking:
 - Sampling - Geometry
 - Scoring - Energy
- Still significant room for improvement
 - Lots of efforts focused on the creation of novel scoring functions
 - Empirical
 - Knowledge based
 - Force field based
- In this presentation
 - Focus on the role of free energy calculations to score molecules





Molecular Mechanics



Molecular Mechanics: use of the potential energy

$$E_{\text{potential}} = E_{\text{bond}} + E_{\text{angle}} + E_{\text{torsion}} + E_{\text{van der Waals}} + E_{\text{coulomb}}$$



Biological Force Fields

- Effective pair potentials
- Simple functional form

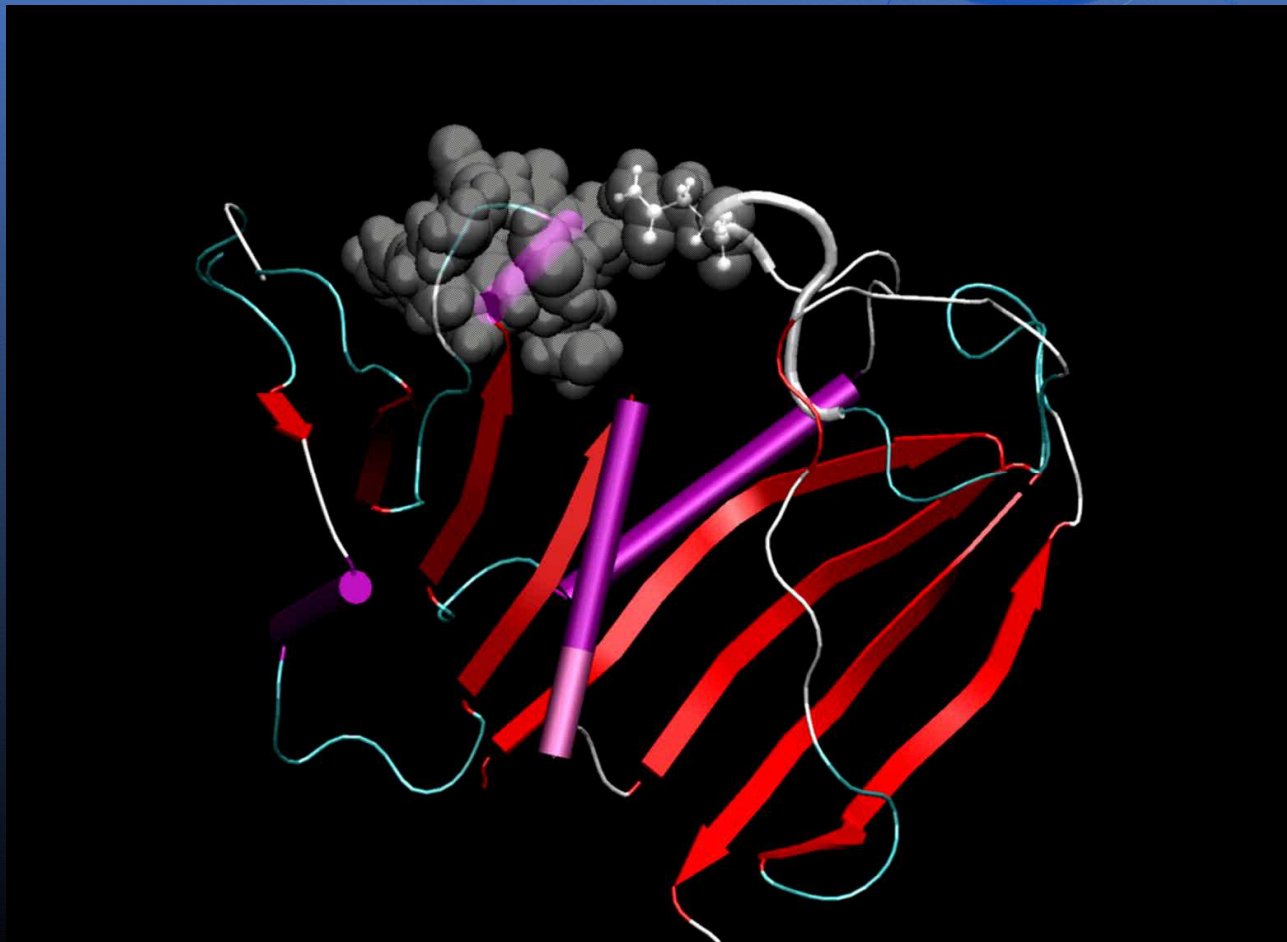
$$U = \sum_{\text{pairs}(i,j)} \frac{q_i q_j}{4\pi\epsilon_0 r_{ij}} + 4\epsilon \left[\left(\frac{\sigma}{r} \right)^{12} - \left(\frac{\sigma}{r} \right)^6 \right]$$

- More complex functional forms generally avoided



DHFR

- Molecular dynamics, with modifications to drive flexible loop





Challenges?

- Computational cost
 - For industrial use, need 10+ compounds per night
- Force field
 - Effective pair potentials
 - Accurate, but only with extensive parameterisation
 - Violates cost criterion!
 - QM/MM?
- Sampling
 - Satisfactory for a particular binding geometry
 - What if this changes though?
 - Novel, efficient sampling algorithms needed
- Applicability
 - Very similar compounds

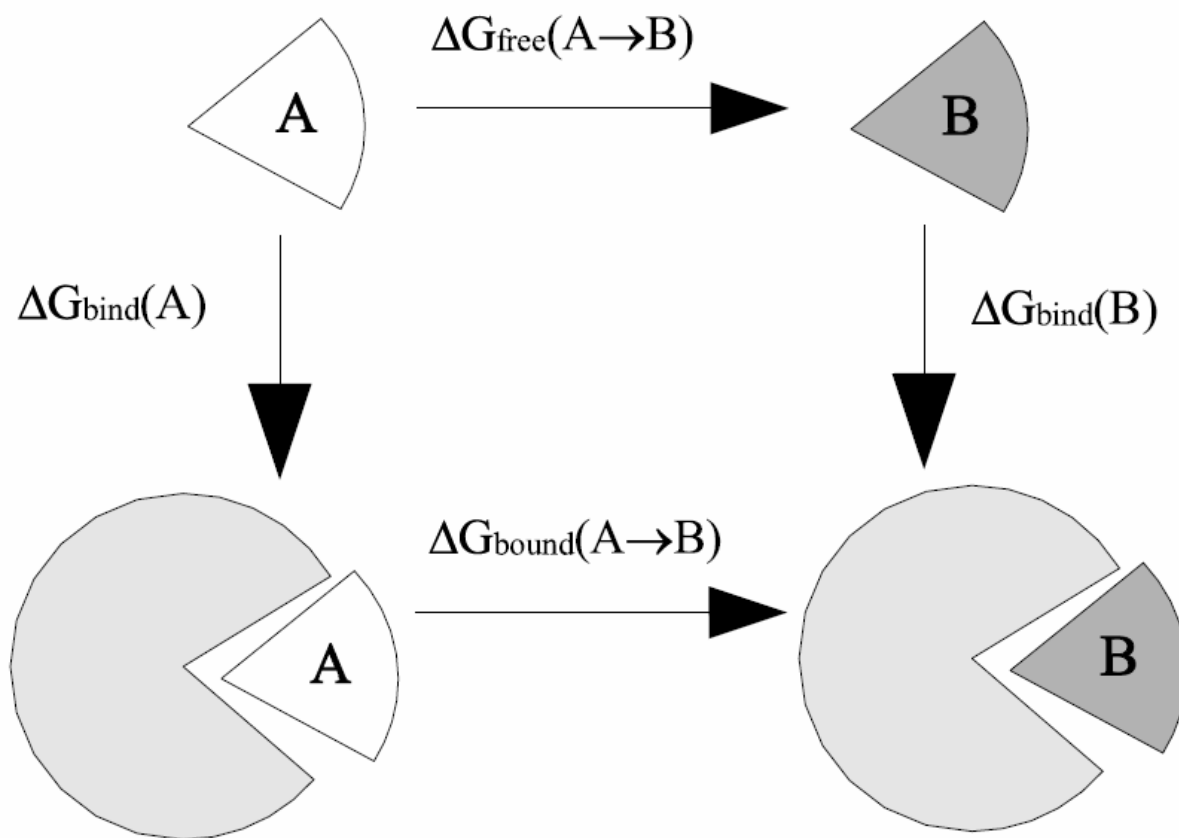


Methodology

- Speed essential
- Statistical thermodynamics
 - Replica exchange
- Monte Carlo sampling
 - Flexible ligands and protein side chains
- Implicit solvent framework
 - GBSA parameterisation
 - Dual potential MC
- Large ligand differences
 - Dual topology



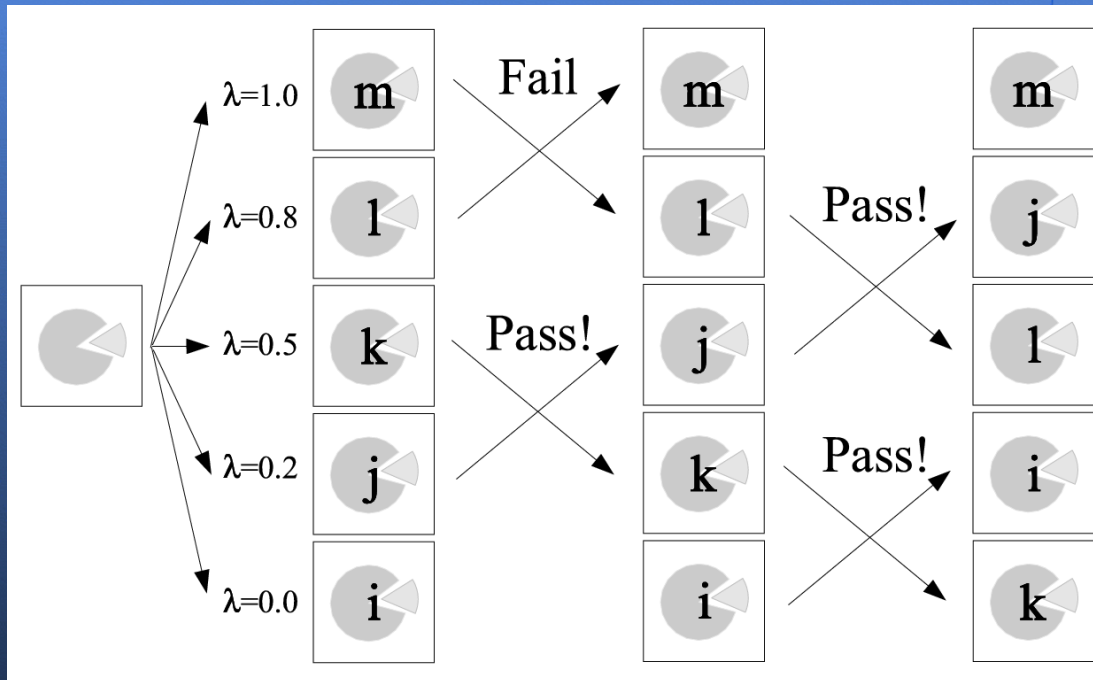
Predicting binding free energies



$$\Delta G_{\text{bind}}(B) - \Delta G_{\text{bind}}(A) = \Delta G_{\text{bound}}(A \rightarrow B) - \Delta G_{\text{free}}(A \rightarrow B)$$



Replica-exchange free energy

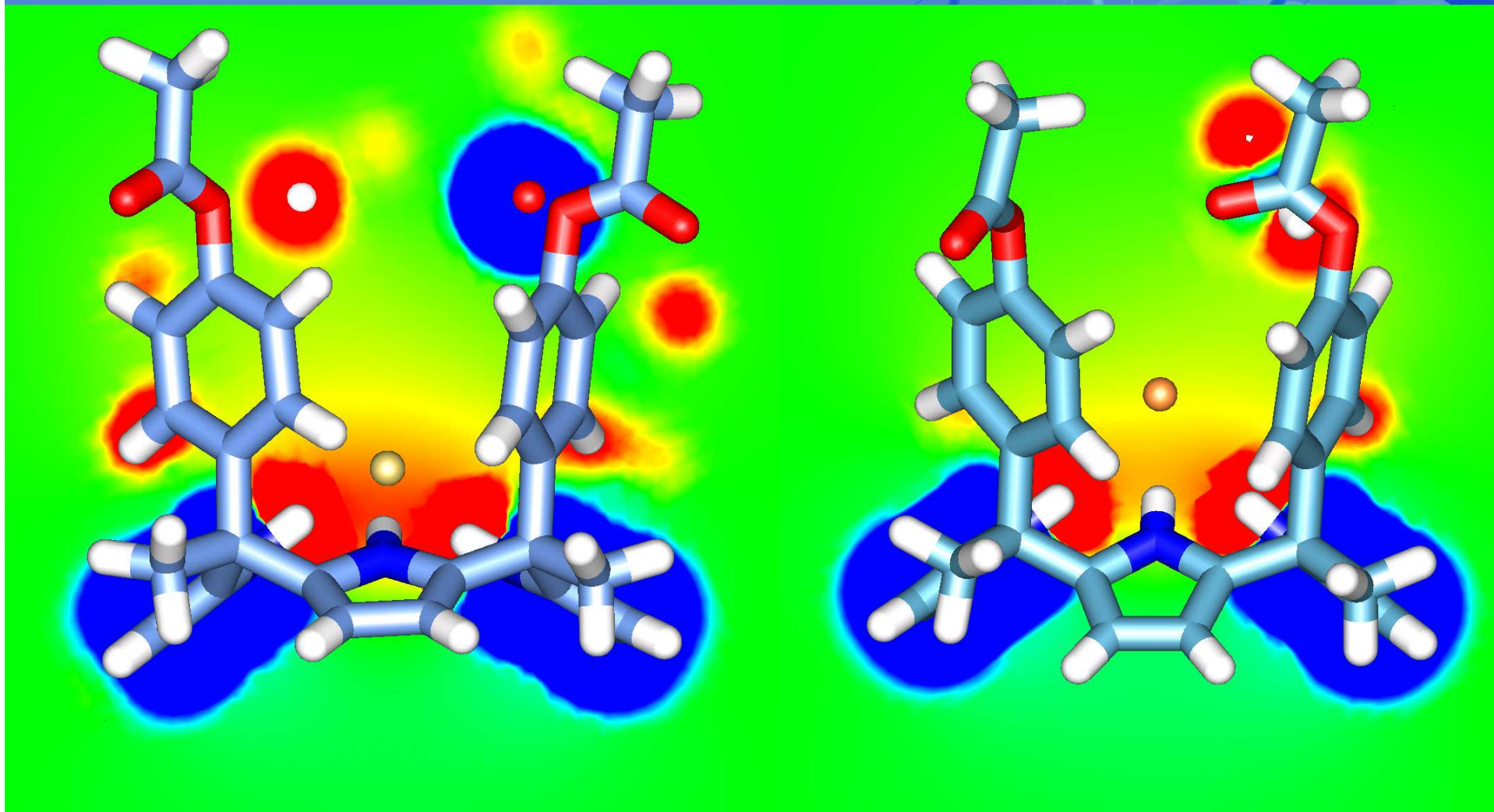


$$\Delta G = \int_{\lambda=0}^{\lambda=1} \left\langle \frac{\partial H}{\partial \lambda} \right\rangle_{\lambda} d\lambda$$

- Free energy methods that are applied between exchanges are the same as normal
- Exchanges require little extra computational cost



Calix[4]pyrrole/halide complexes

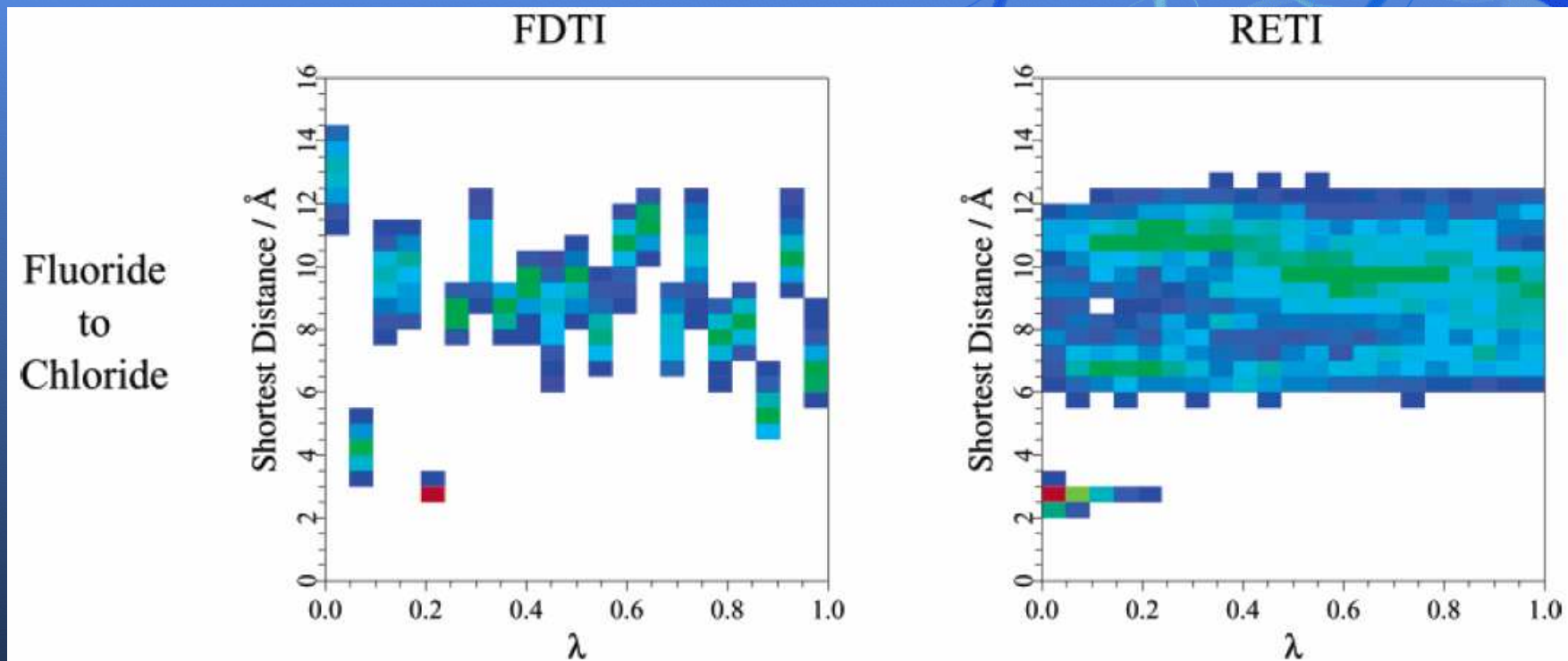


Fluoride bound structure

Chloride bound structure

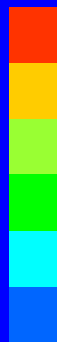


Distances to contaminating water



Fluoride
to
Chloride

Density



High

Medium

Low

RETI has weighted
the available
configurations
across λ

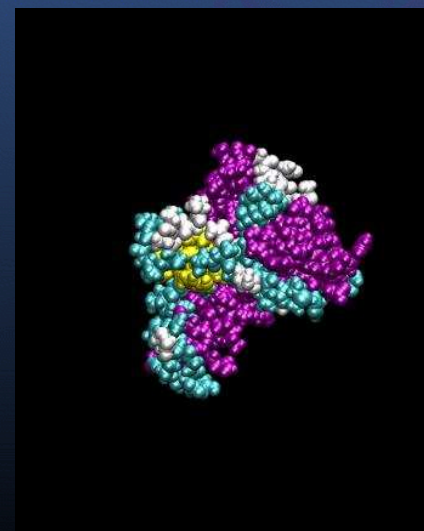
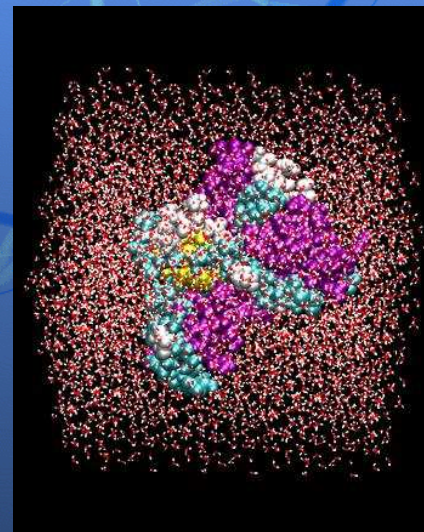
J. Phys. Chem. B 107,
13703-13710, **2003**

J. Phys. Chem. B 107,
13711-13718, **2003**



Implicit solvent model

- Generalised Born
 - Pairwise Descreening Approximation
- Parameterisation
 - Dataset of small molecules
AMBER/AM1-BCC
 - Optimisation
Genetic Algorithm
 - Validation
Cross-validation
Potential of Mean Force calculations
- Approximations



J. Comput. Chem. 25, 1760-1770, 2004

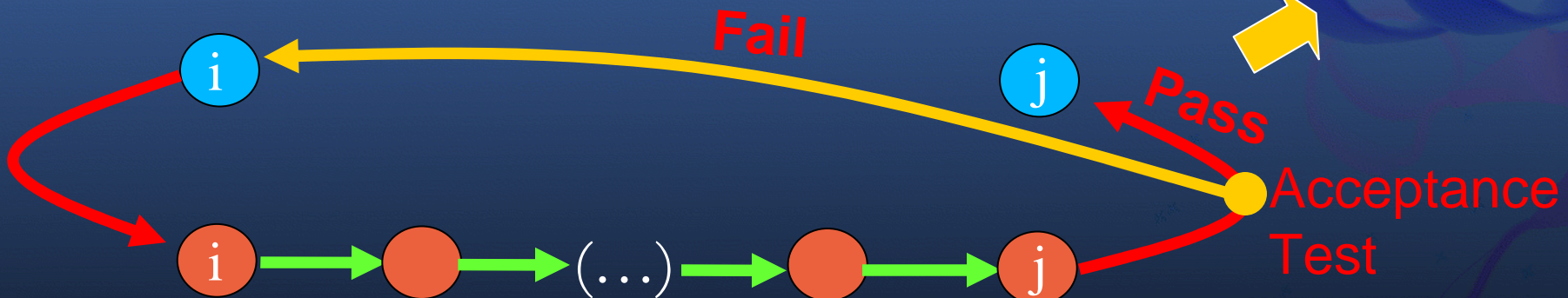


Simplified sampling potential

- Metropolis Monte Carlo : Random Walk



- Random Walk with two potentials



- Sample N steps with the **red** potential
- Add state **j** to the **blue** ensemble with probability

$$\chi = (\pi_j/\pi_i) * (\pi_i/\pi_j)$$



Simplified sampling potential: Application

- Reference Potential
 - $\text{GBSA}_{\text{approx}}$
 - GBSA with SA and approximated GB
- Two simplified sampling potentials
 - DDD
 - Distance Dependent Dielectric Model (no SA)
 - FastGB
 - GB model with low cutoffs (no SA)

FastGB

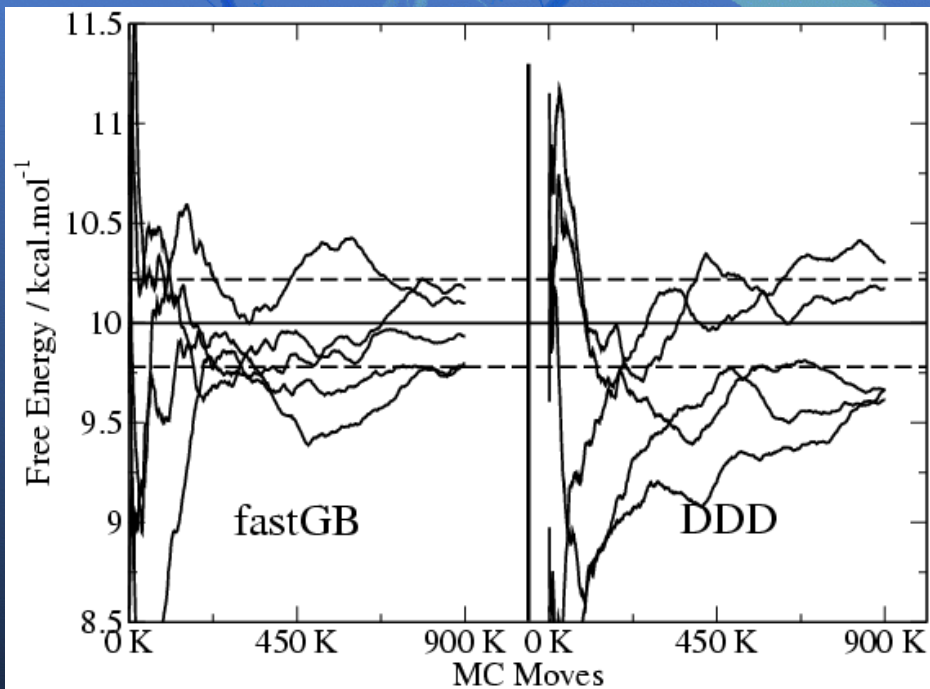
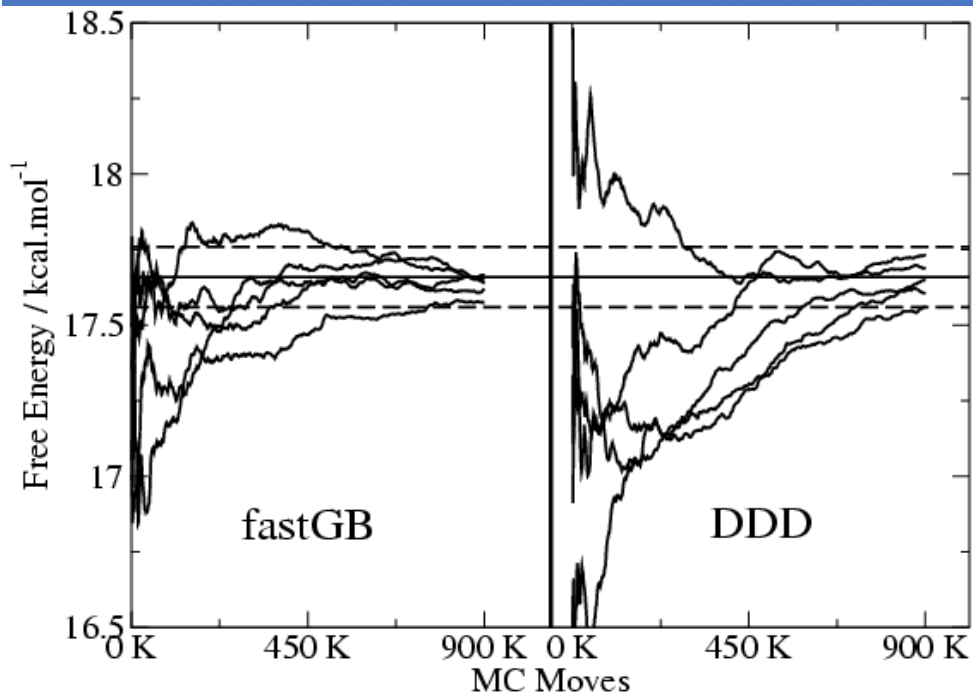
Speed up **x2.3**
Acceptance rate
at $\chi = 85\%$

DDD

Speed up **x2.5**
Acceptance rate
at $\chi = 42\%$



Simplified sampling potential: Results

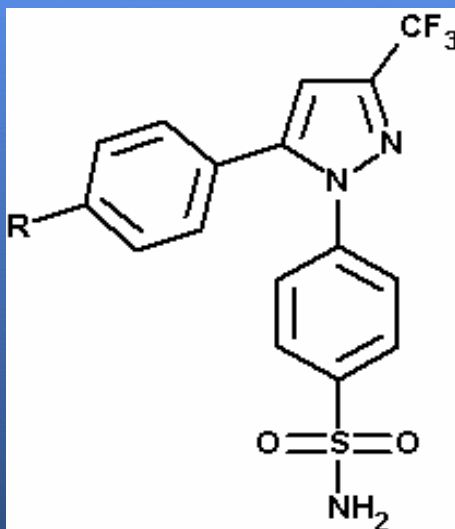


- Faster convergence of fastGB/GBSA_{approx} over DDD/GBSA_{approx}

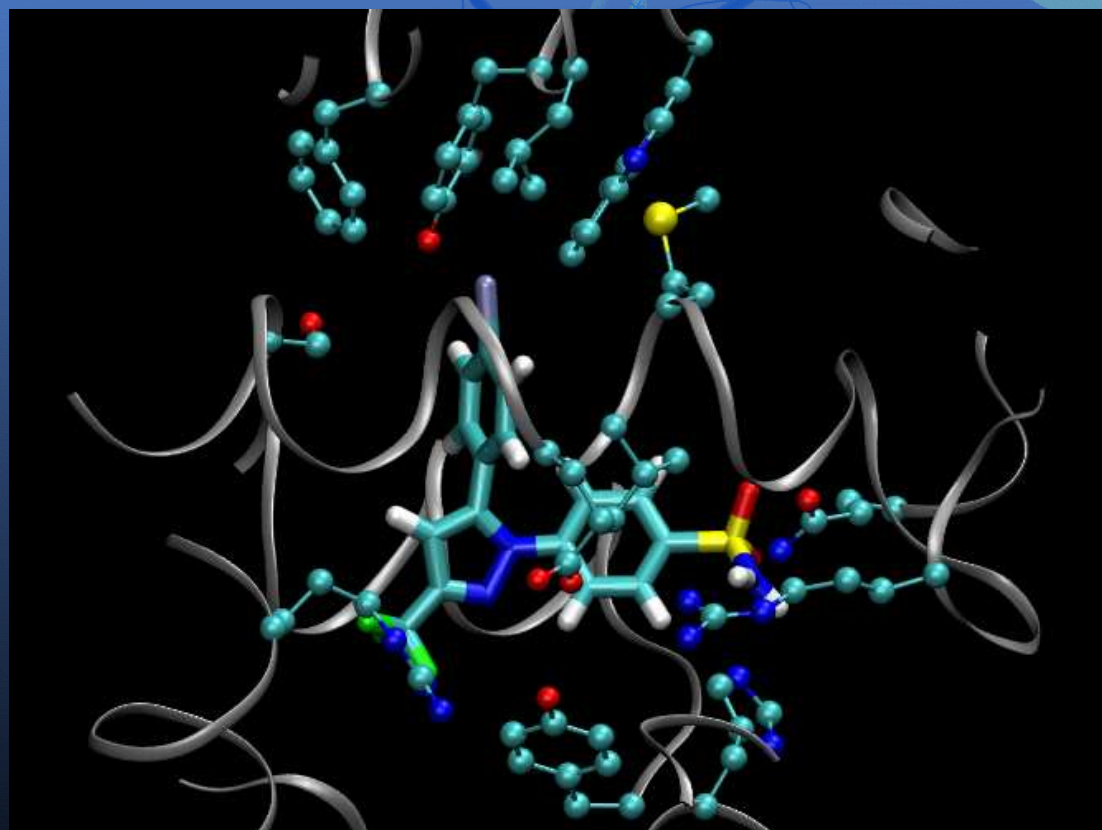
J. Chem. Theory Comput. 2, 732-739, 2006



Case study: COX2

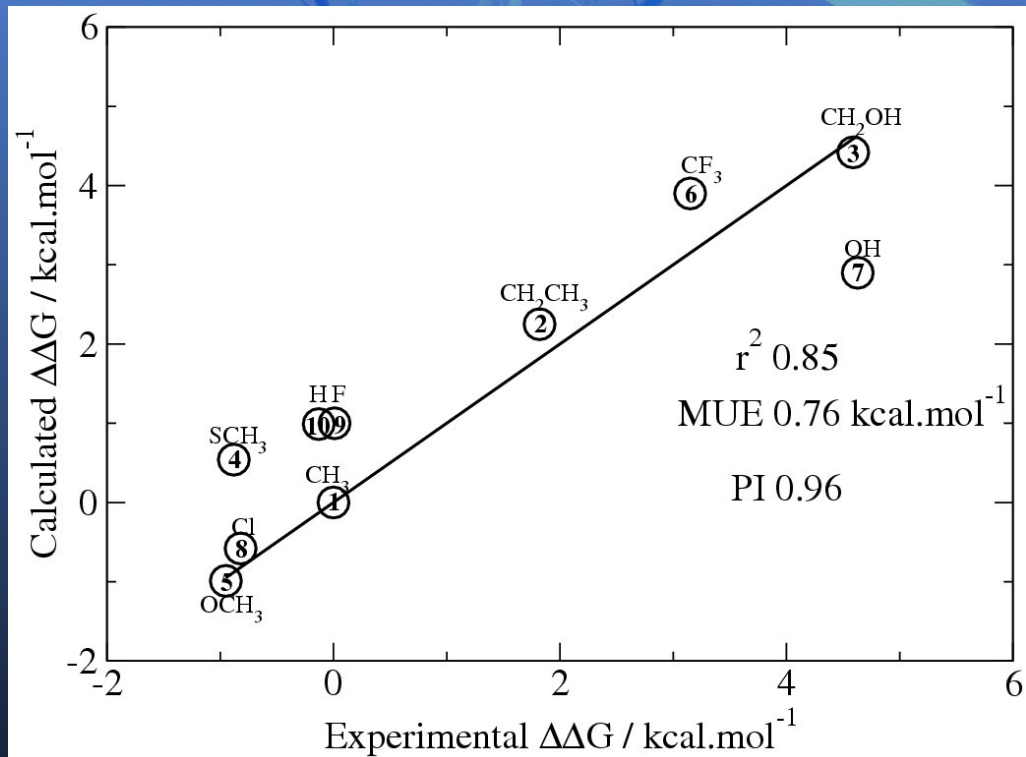
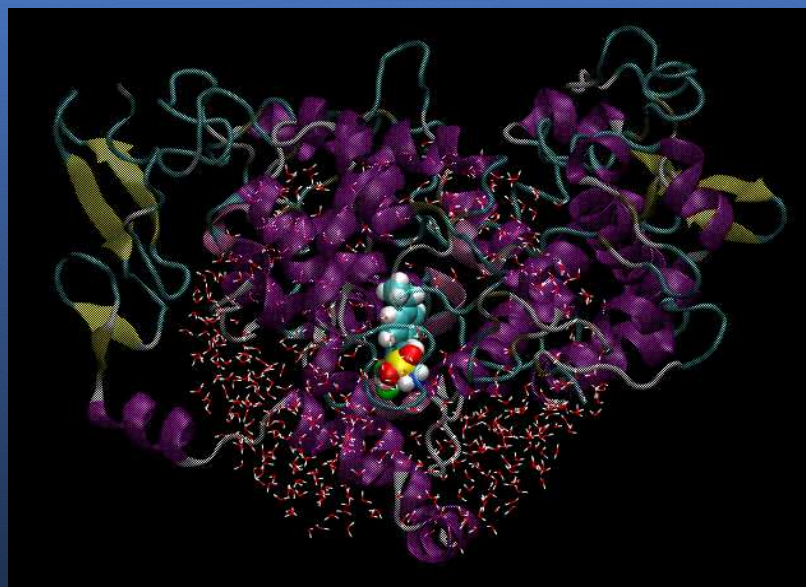


Compound	R	IC ₅₀ (μM)
1	CH ₃	0.04
2	CH ₂ CH ₃	0.86
3	CH ₂ OH	93.3
4	SCH ₃	0.009
5	OCH ₃	0.008
6	CF ₃	8.23
7	OH	>100
8	Cl	0.01
9	F	0.041
10	H	0.032





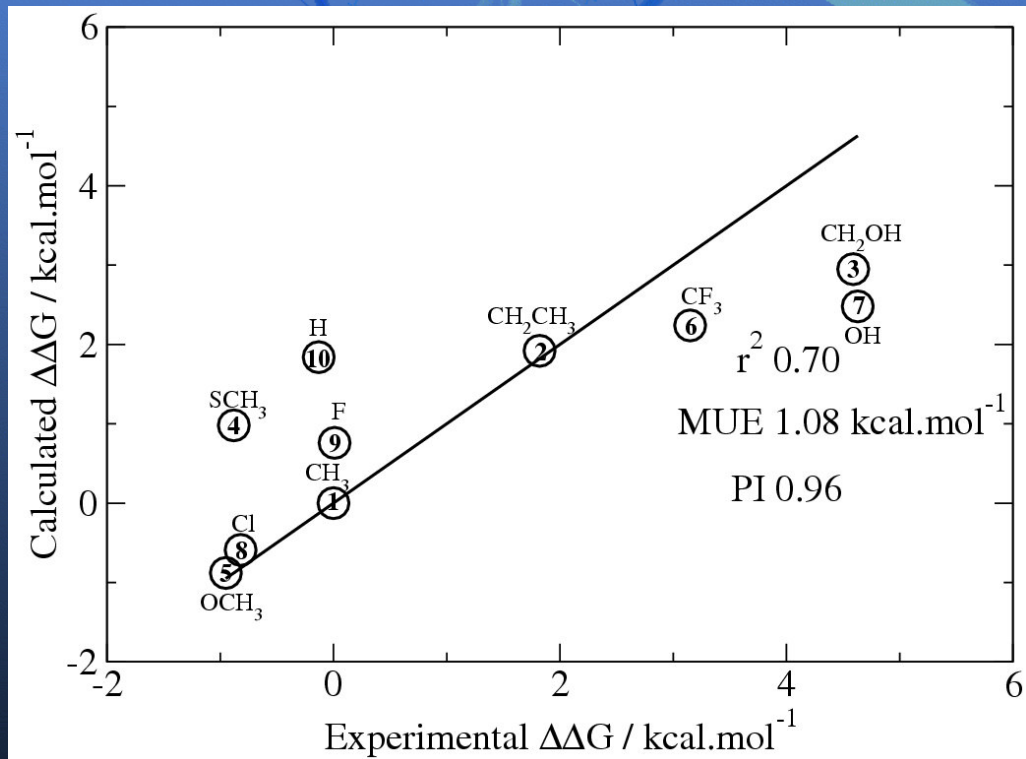
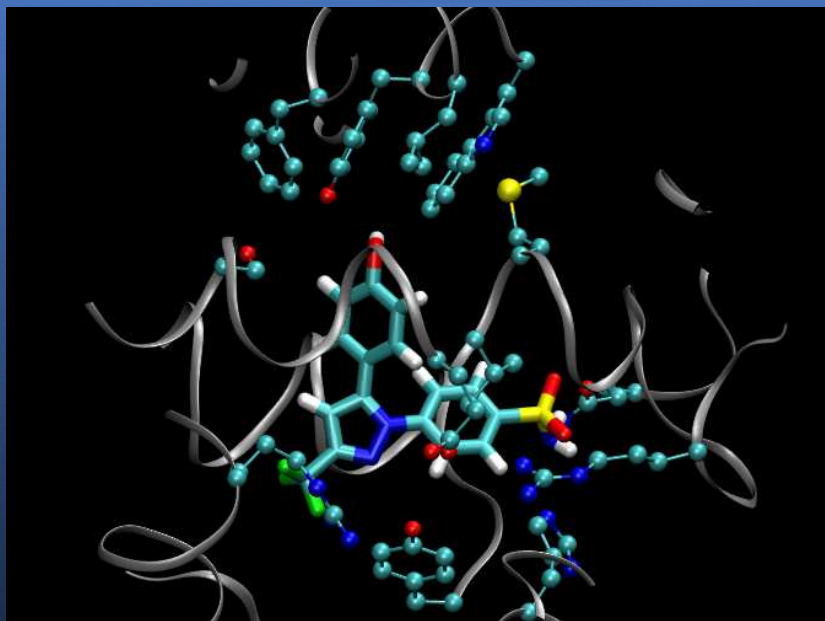
COX2: Explicit solvent results



PI = Predictive index: quality of rank ordering

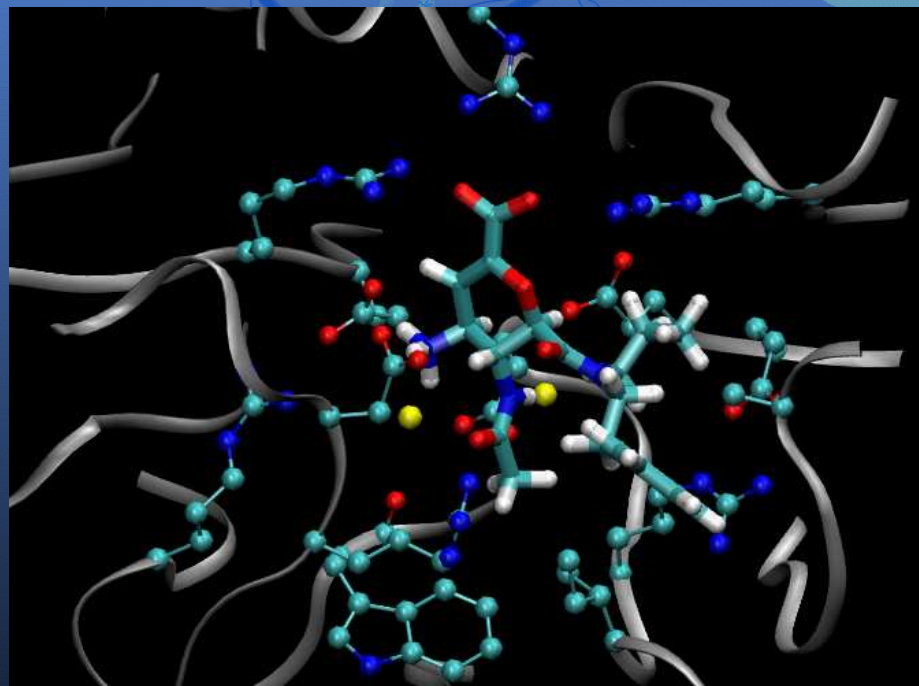
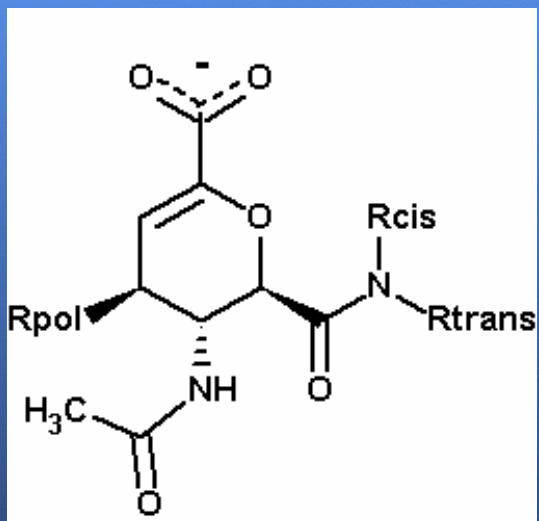


COX2: Implicit solvent results





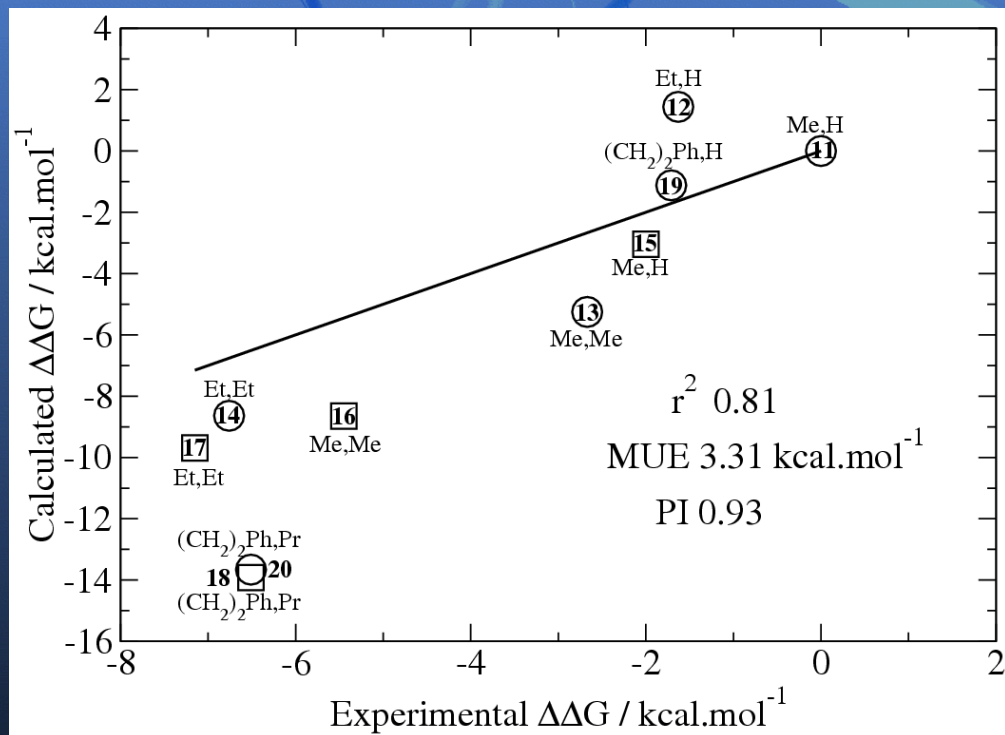
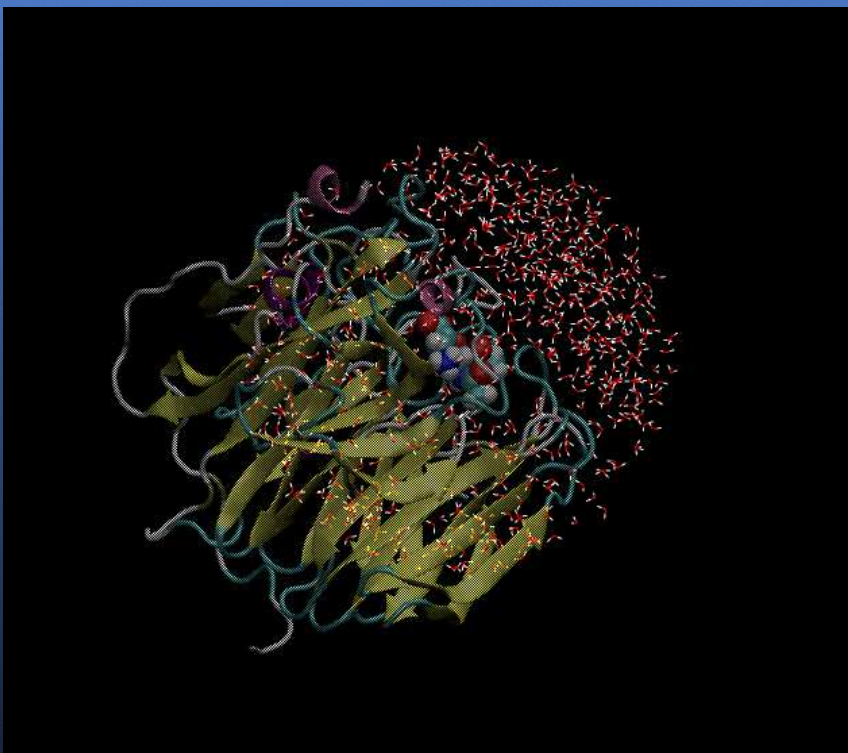
Case study: Neuramininidase



Compound	R _{trans}	R _{cis}	R _{pol}	IC ₅₀ (μM)
11	Me	H	NH ₃ ⁺	190
12	Et	H	NH ₃ ⁺	13
13	Me	Me	NH ₃ ⁺	2.4
14	Et	Et	NH ₃ ⁺	0.003
15	Me	H	NH ₃ ⁺	7
16	Me	Me	NHC(NH ₂) ₂ ⁺	0.025
17	Et	Et	NHC(NH ₂) ₂ ⁺	0.001
18	(CH ₂) ₂ Ph	Pr	NHC(NH ₂) ₂ ⁺	0.005
19	(CH ₂) ₂ Ph	H	NHC(NH ₂) ₂ ⁺	12
20	(CH ₂) ₂ Ph	Pr	NHC(NH ₂) ₂ ⁺	0.005

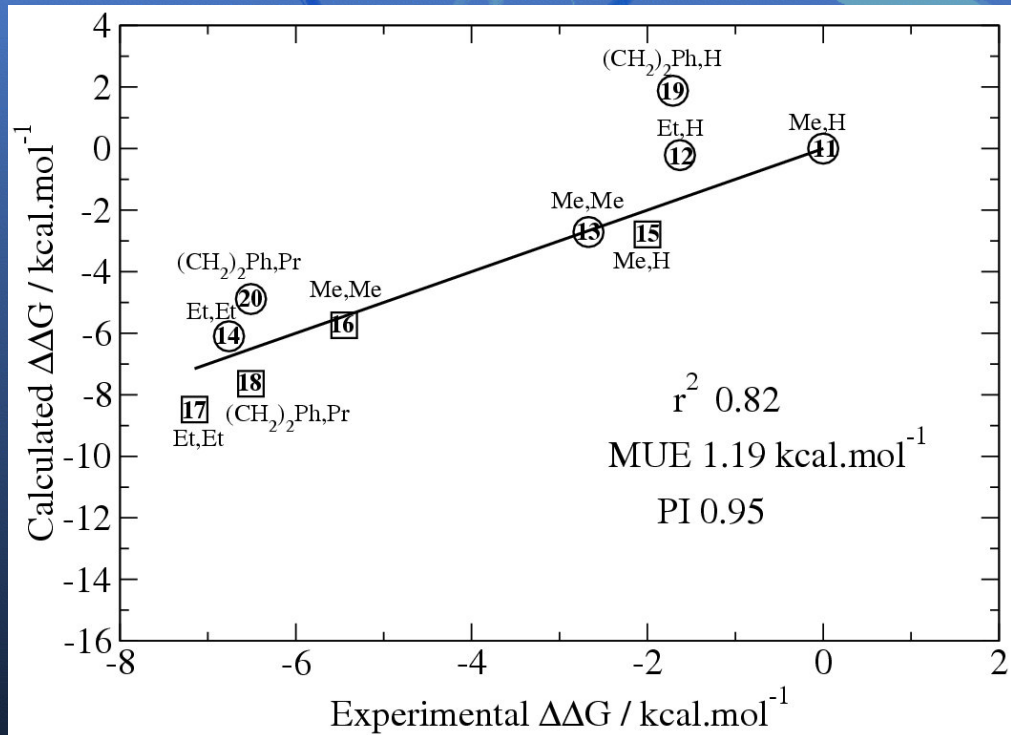
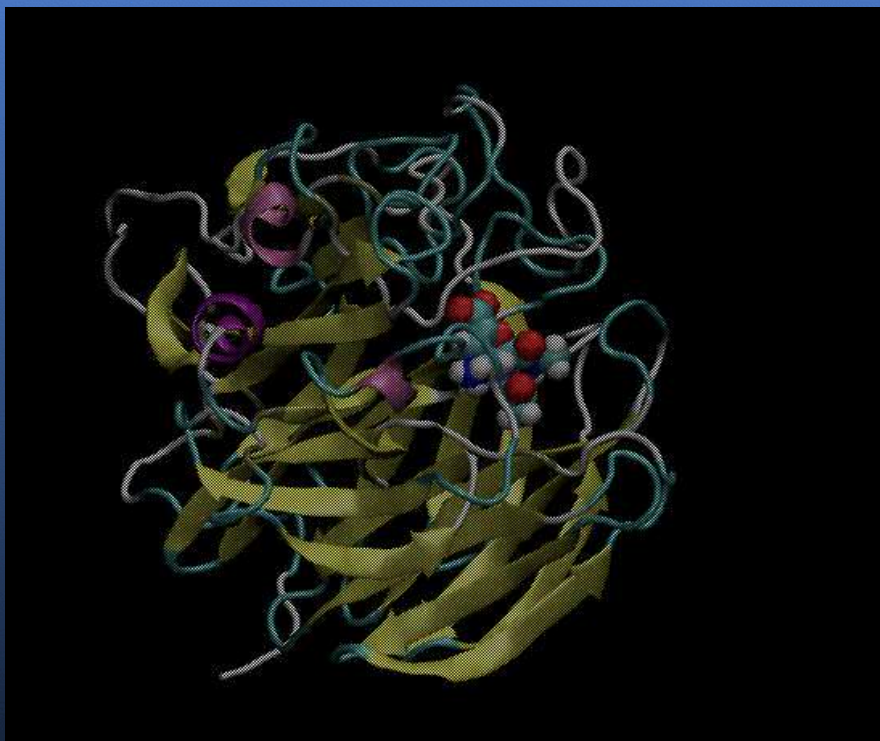


Neuraminidase: Explicit solvent results





Neuraminidase: Implicit solvent results



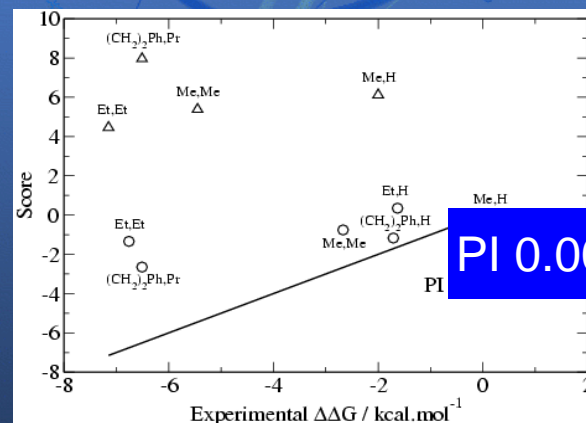
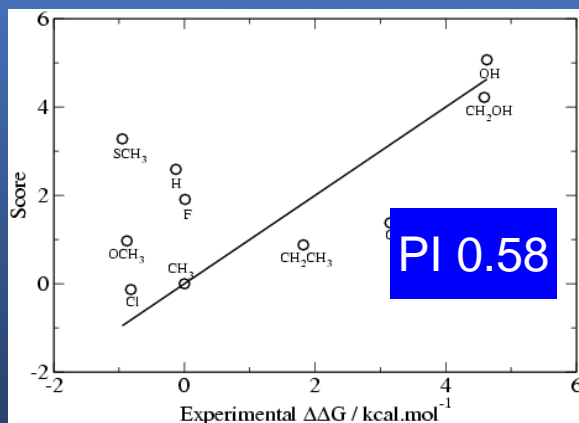


Ranking: Empirical scoring functions

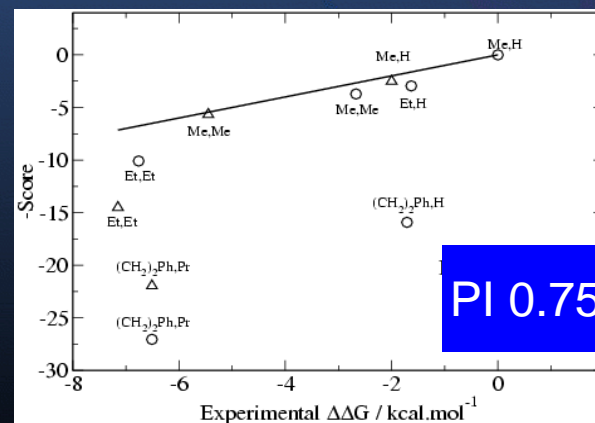
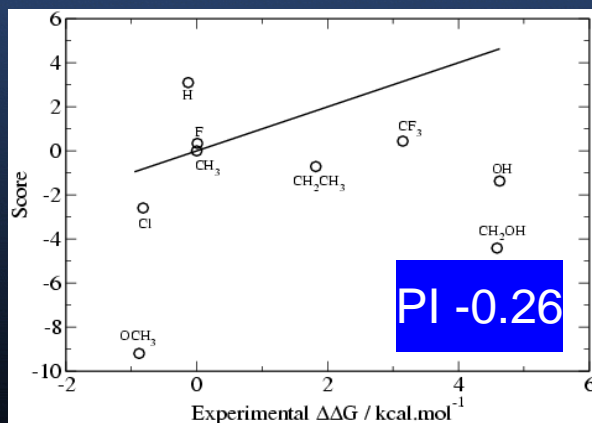
COX2

Neuraminidase

Chemscore



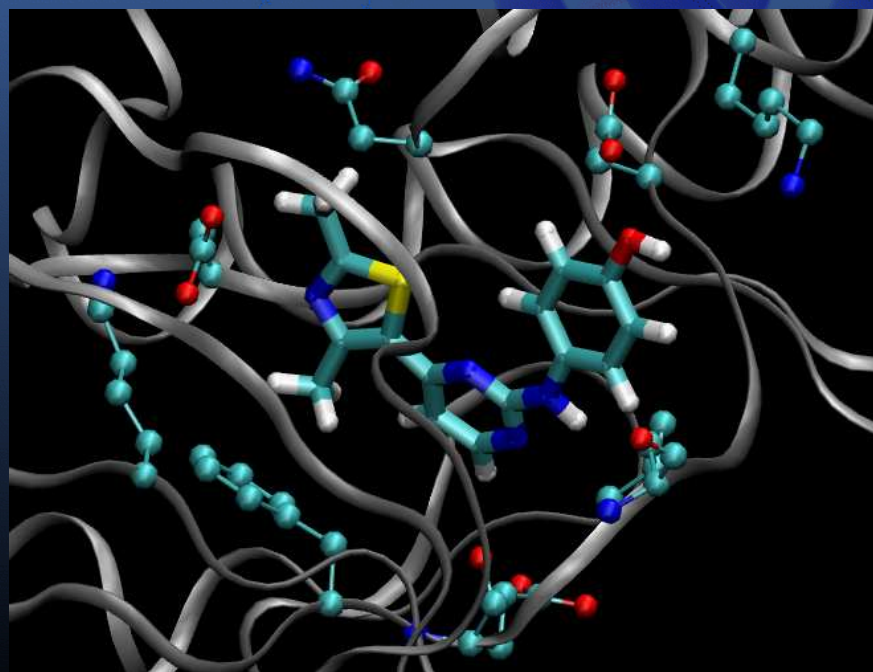
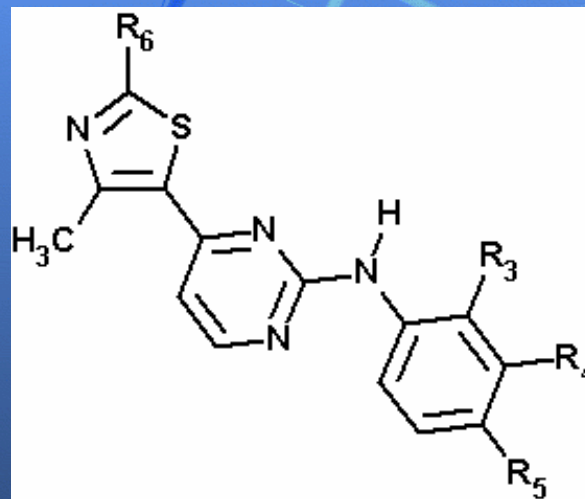
Goldscore





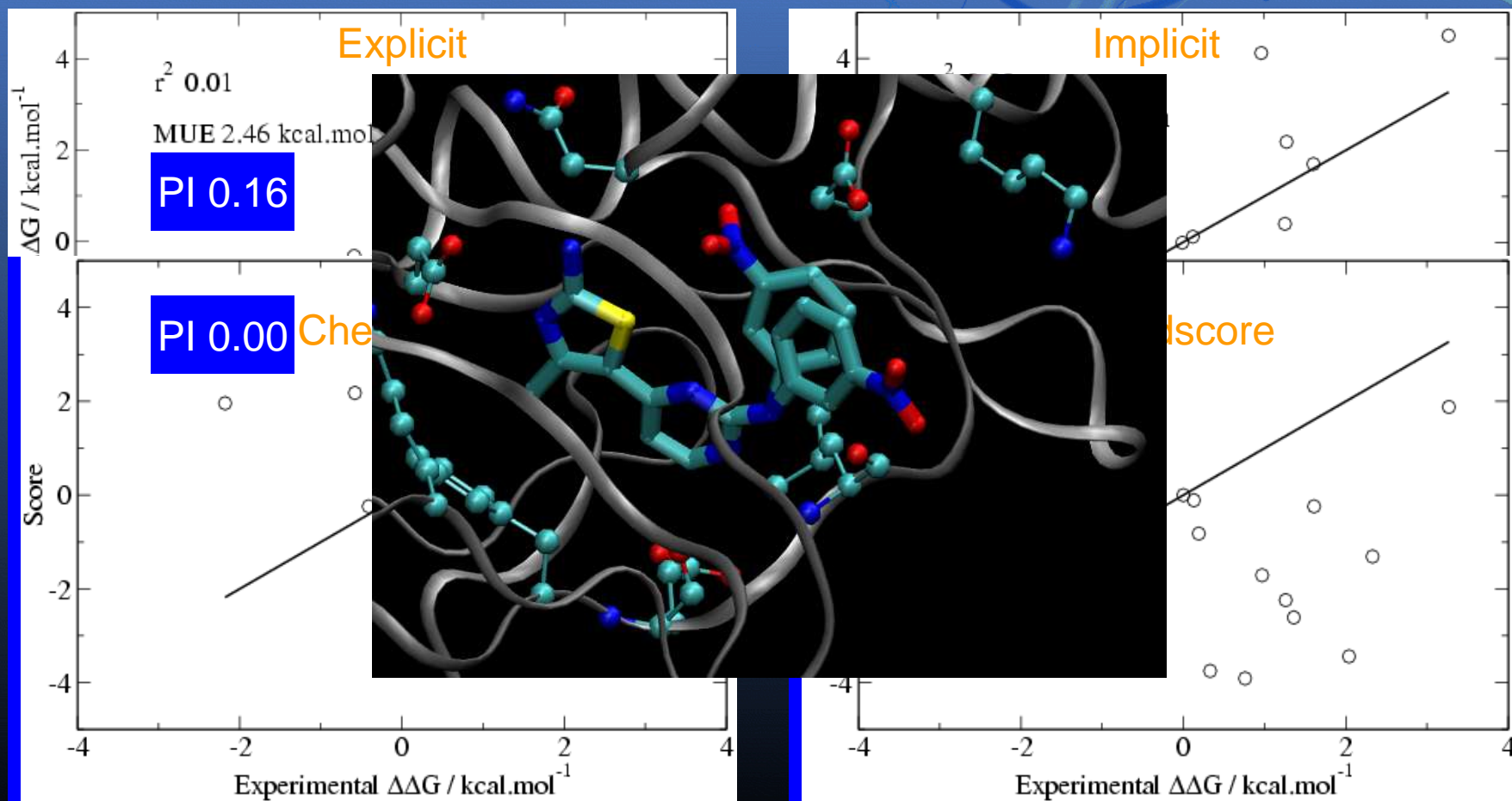
Case study: CDK2

Compound	R ₃	R ₄	R ₅	R ₆	IC ₅₀ (μM)
21	H	H	H	Me	0.08
22	Cl	H	H	Me	>20
23	H	Cl	H	Me	0.67
24	H	H	Cl	Me	2.5
25	F	H	H	Me	1.2
26	H	F	H	Me	0.1
27	H	H	F	Me	0.04
28	H	H	CF ₃	Me	0.29
29	H	OH	H	Me	0.06
30	H	H	OH	Me	0.14
31	H	H	OH	NHMe	0.07
32	H	H	OH	NH ₂	0.03
33	H	NO ₂	H	Me	0.11
34	H	NO ₂	H	NHMe	0.8
35	H	NO ₂	H	NH ₂	0.002
36	H	H	NO ₂	Me	4.1
37	H	NH ₂	H	Me	0.4
38	H	H	H	NMe ₂	0.7





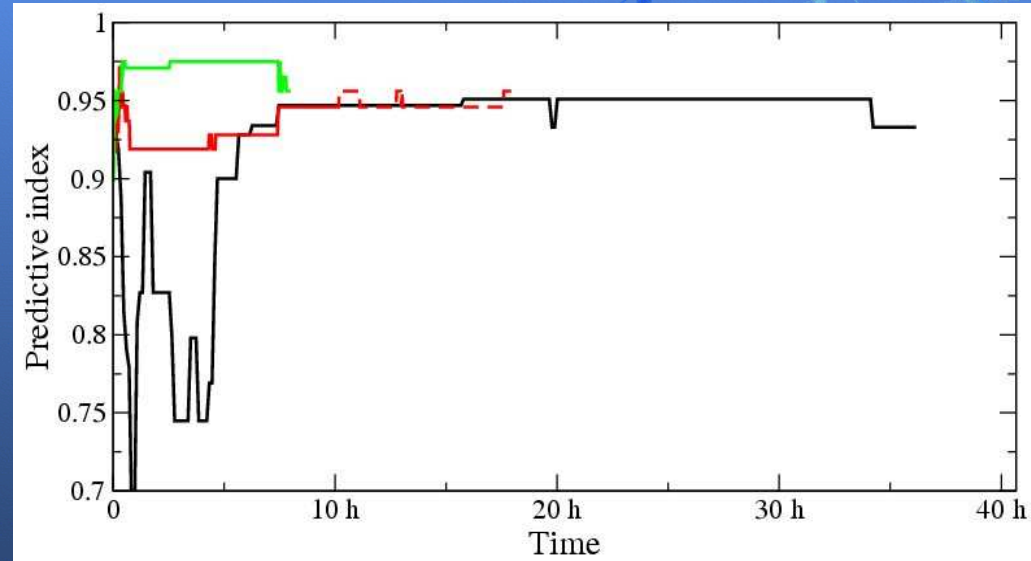
CDK2: Results



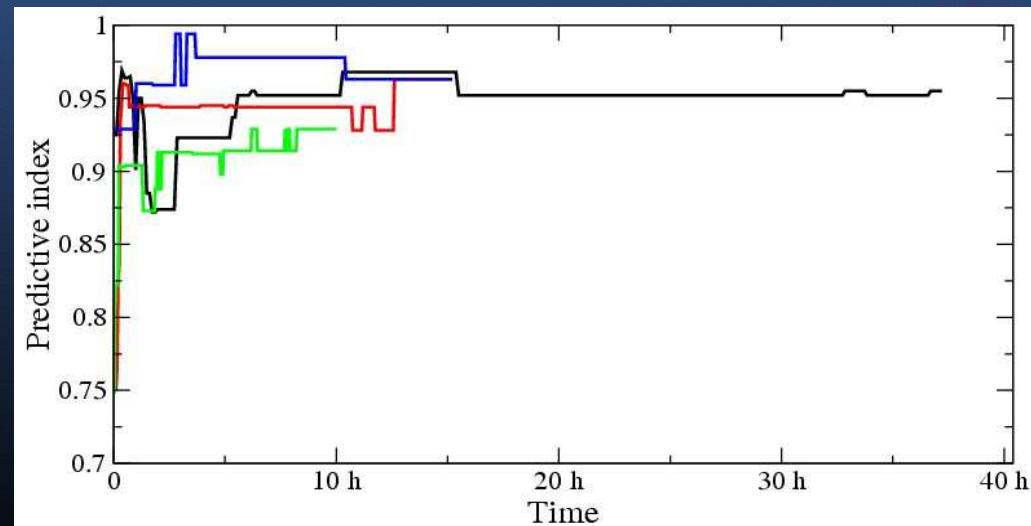


Convergence

Neuraminidase

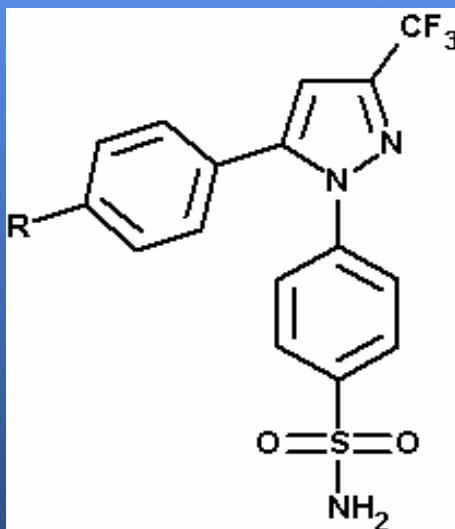


COX2





Free energy calculations in drug design



J. Med. Chem., 49, 7427-7439, 2006

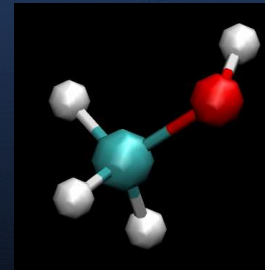
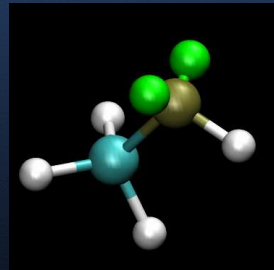
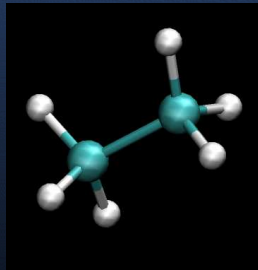
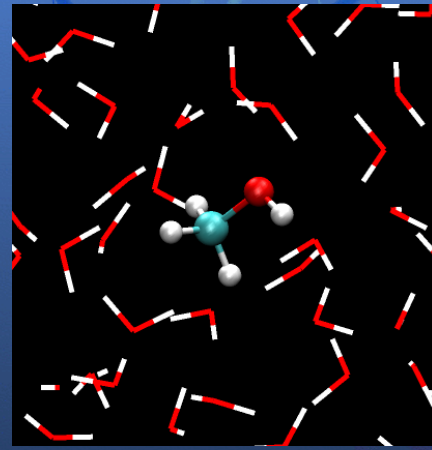
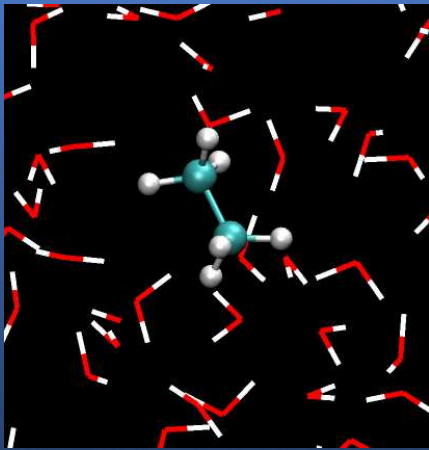
- Restricted to similar (congeneric) ligands
- Fast, but fast enough?

A general methodology that can handle structurally diverse ligands would be very useful

Compound	R	IC ₅₀ (μM)
1	CH ₃	0.04
2	CH ₂ CH ₃	0.86
3	CH ₂ OH	93.3
4	SCH ₃	0.009
5	OCH ₃	0.008
6	CF ₃	8.23
7	OH	>100
8	Cl	0.01
9	F	0.041
10	H	0.032



Ethane to methanol: single topology



0.0

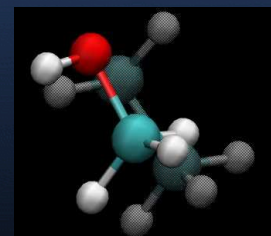
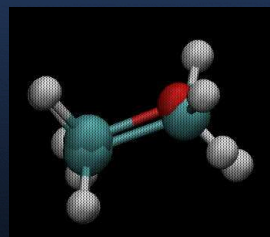
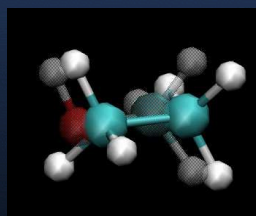
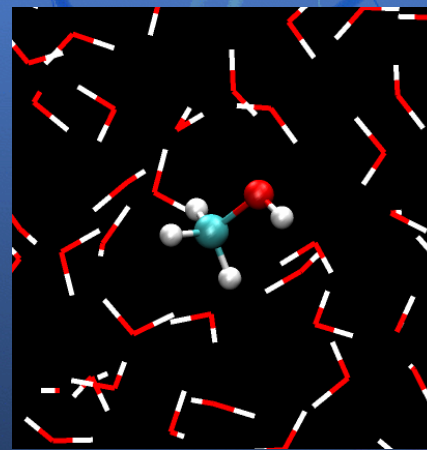
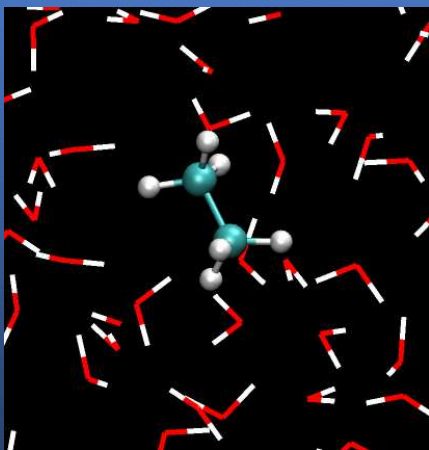
λ

1.0





Ethane to methanol: dual topology



0.0

λ

1.0



Dual topology: implementation

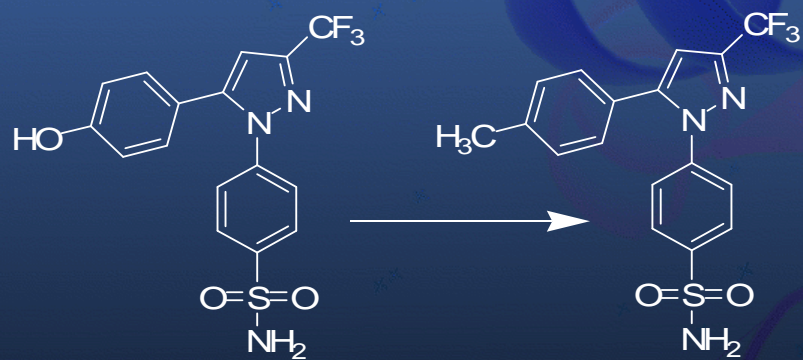
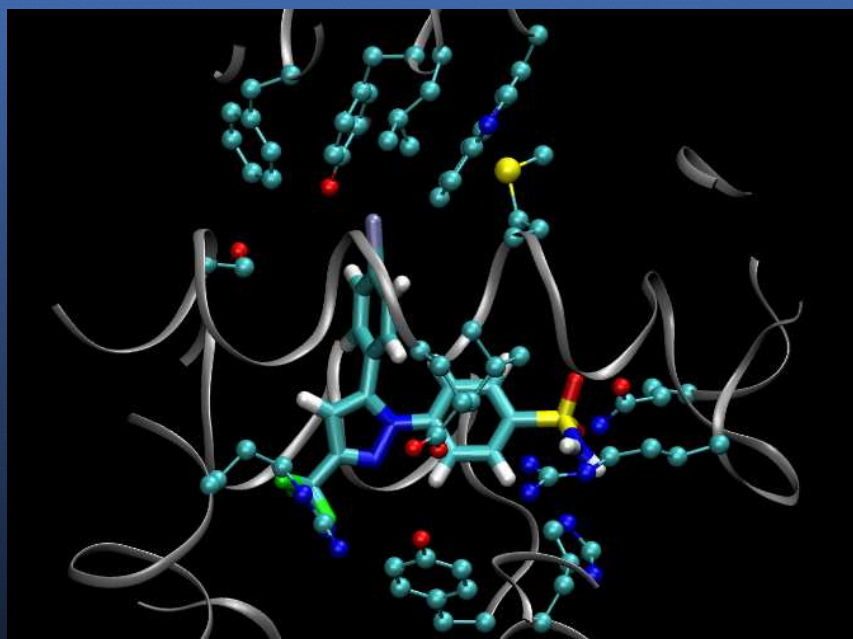
- Complete dual topology
 - Perturb molecules with no common structure
- Flexible softcore energy function

$$U_{nonbonded,\lambda} = (1-\lambda)4\epsilon_{ij} \left[\left(\frac{\sigma_{ij}^{12}}{(\lambda\delta\sigma_{ij} + r_{ij}^2)^6} \right) - \left(\frac{\sigma_{ij}^6}{(\lambda\delta\sigma_{ij} + r_{ij}^2)^3} \right) \right] + \frac{(1-\lambda)^n q_i q_j}{4\pi\epsilon_0 \sqrt{(\lambda + r_{ij}^2)}}$$

- Coupling of solutes' translation and rotation
 - Vanishing solute does not drift away
- Intramolecular terms not coupled
 - Decoupled solute is transferred to the gas phase

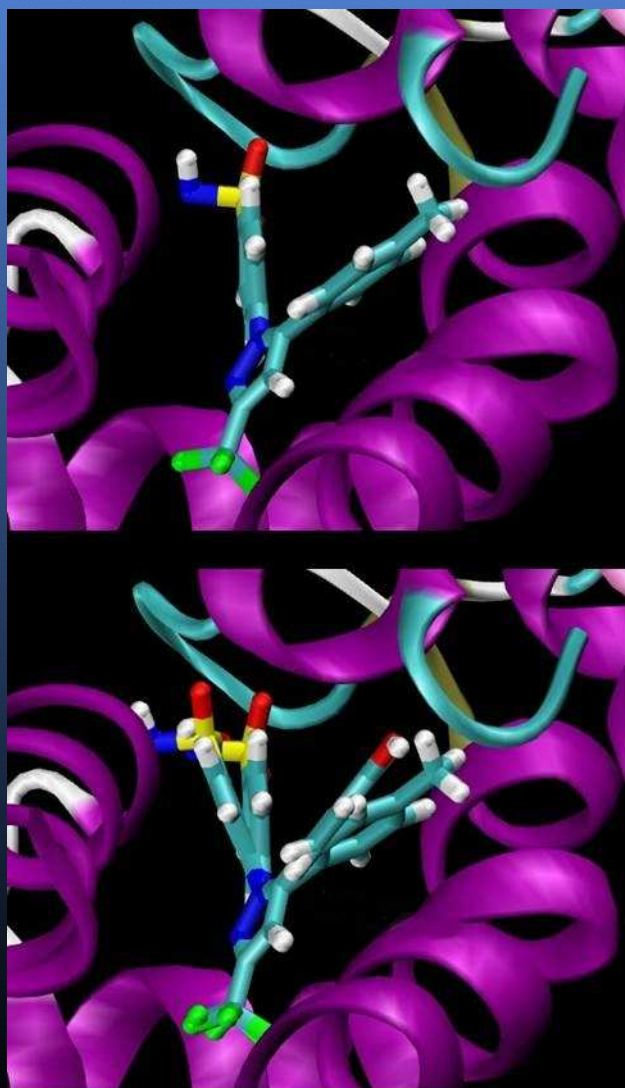


Congeneric inhibitors of COX2





COX2 inhibitors: results



$\Delta\Delta G_{\text{bind}}$ in kcal.mol⁻¹

Protocol	$\langle\Delta\Delta G\rangle$	S_E
dual topology	-2.7	0.6
single topology	-3.0	0.1

$\Delta\Delta G_{\text{solv}}$ in kcal.mol⁻¹

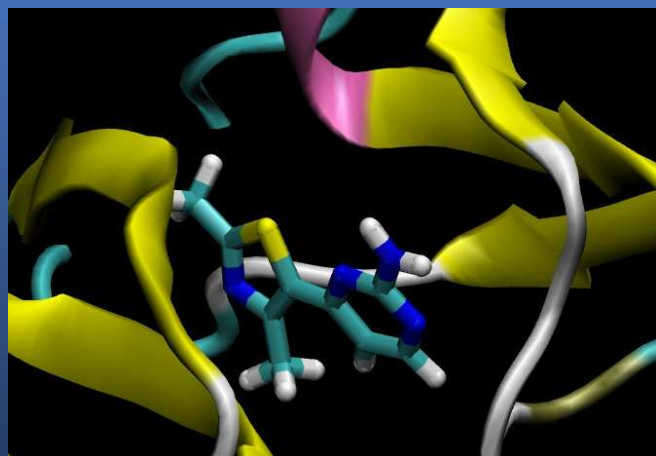
Protocol	$\langle\Delta\Delta G\rangle$	S_E
dual topology	4.6	0.4
single topology	4.5	0.1

- Single topology simulations more precise
- When both methods are applicable, single topology preferred over dual topology

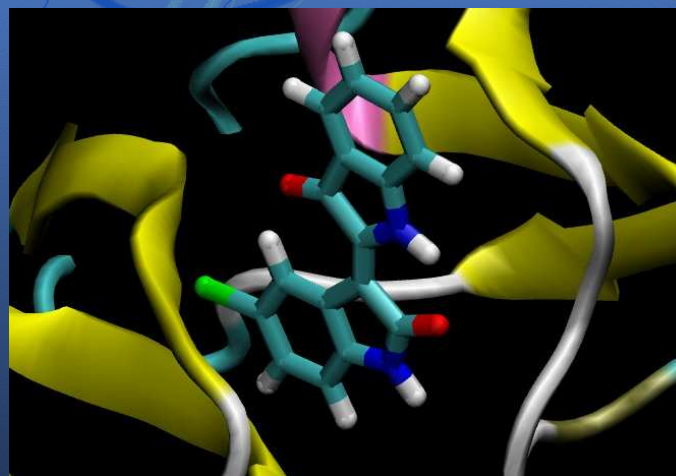


Two CDK2 scaffolds

activated CDK2 / CK hit



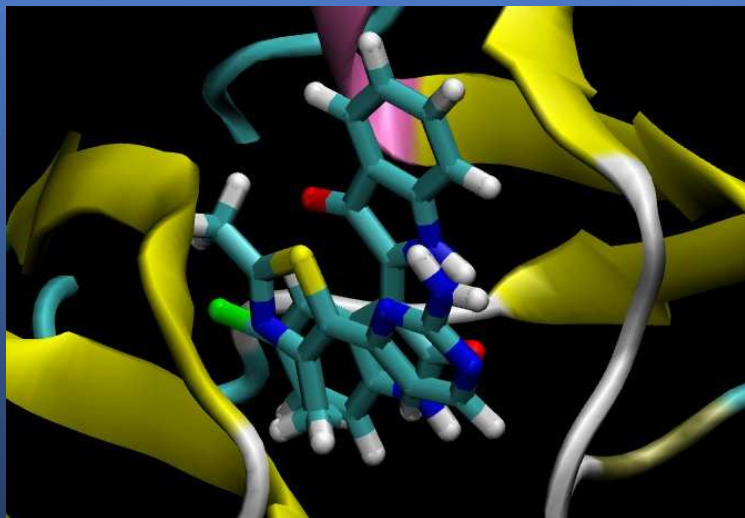
activated CDK2 / 5-bromoindirubine



- No common structural features
- Very difficult to handle by a single topology approach



$\Delta\Delta G$ of two CDK2 scaffolds: results



$$\Delta\Delta G_{\text{bind}} / \text{kcal.mol}^{-1}$$

Protocol	$\langle\Delta\Delta G\rangle$	σ
in TIP4P	-0.48	1.12
in GBSA	-5.62	0.21

$$\Delta\Delta G_{\text{solv}} / \text{kcal.mol}^{-1}$$

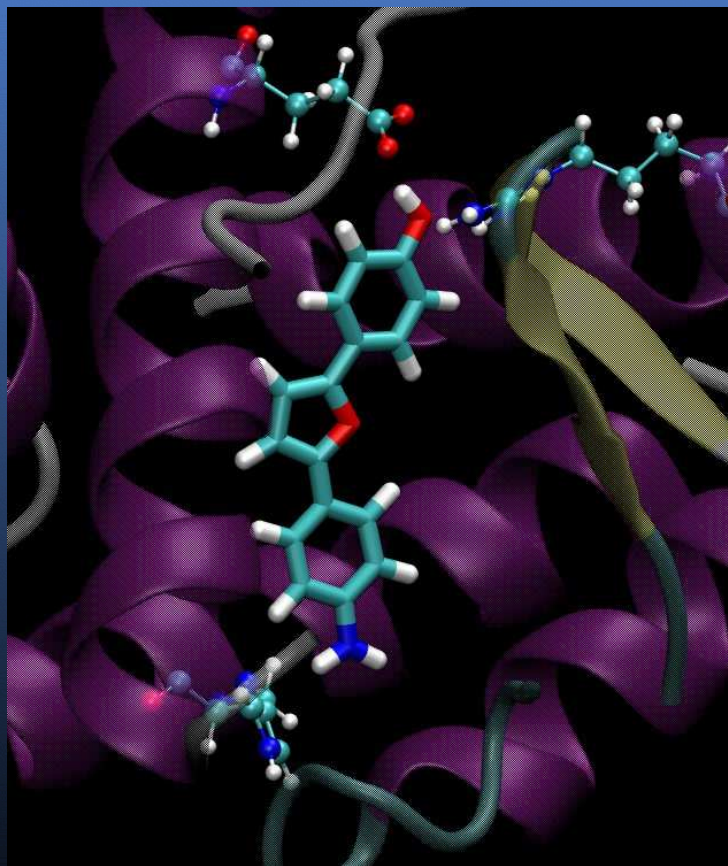
Protocol	$\langle\Delta\Delta G\rangle$	σ
in TIP4P	3.07	0.68
in GBSA	7.14	0.01

- Experiment?
- Implicit solvent simulations more precise
- Sampling and force field

J. Chem. Theory Comput., 3, 1645-1655, 2007



Case study: the estrogen receptor α

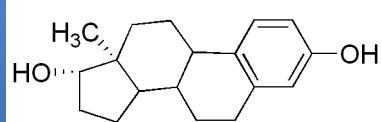


Aims

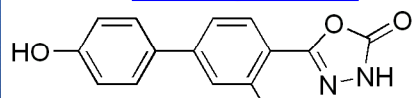
- Test the methodology
- Demonstrate **if**, **how** and **when** free energy simulation techniques can complement existing modelling tools for drug design purposes



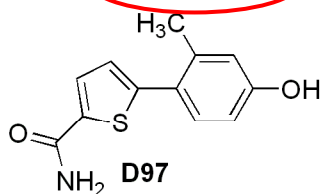
Estrogen receptor α : Ligands



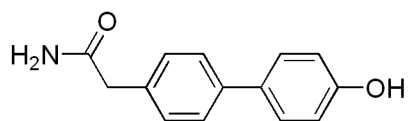
EST, 0.02 μM



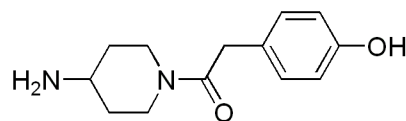
H95, 25 μM



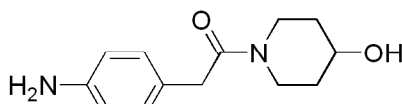
D97



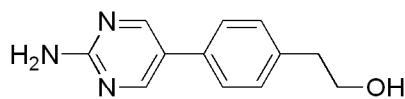
D99



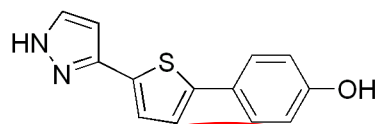
D94



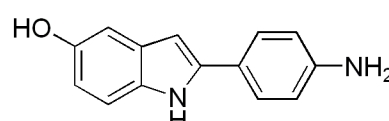
D96



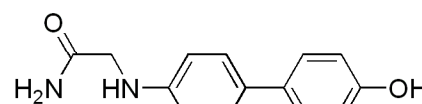
D98



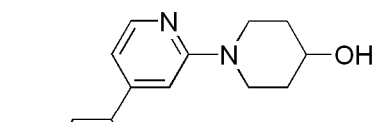
H00, 24 μM



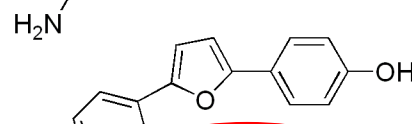
D01



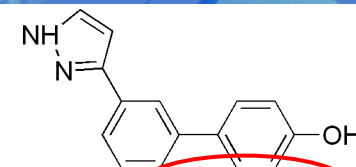
D05



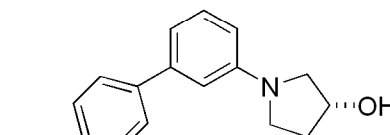
D07



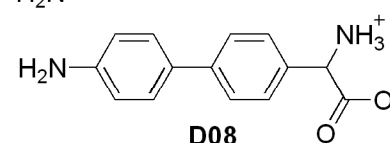
H09, 4.1 μM



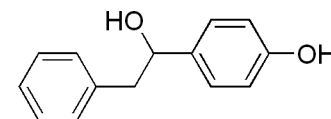
H02, 18 μM



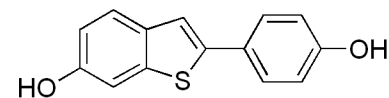
D06



D08



D11

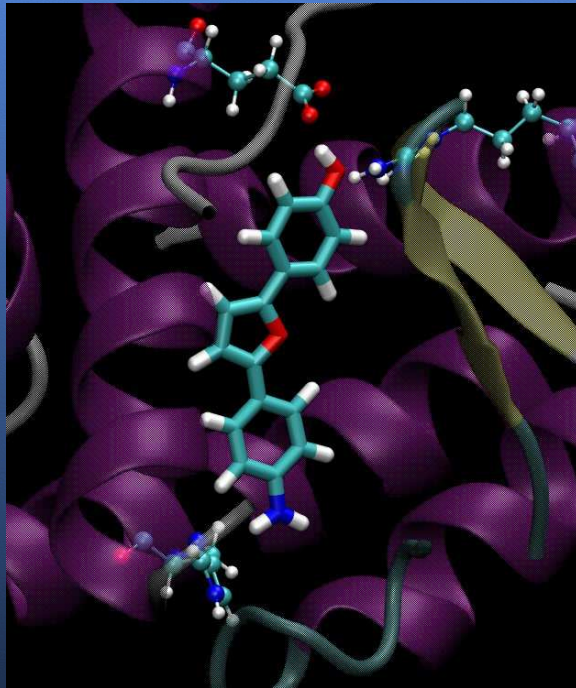


H13, 0.34 μM

Firth-Clark et al., J. Chem. Inf. Model. 46, 642-647, 2006

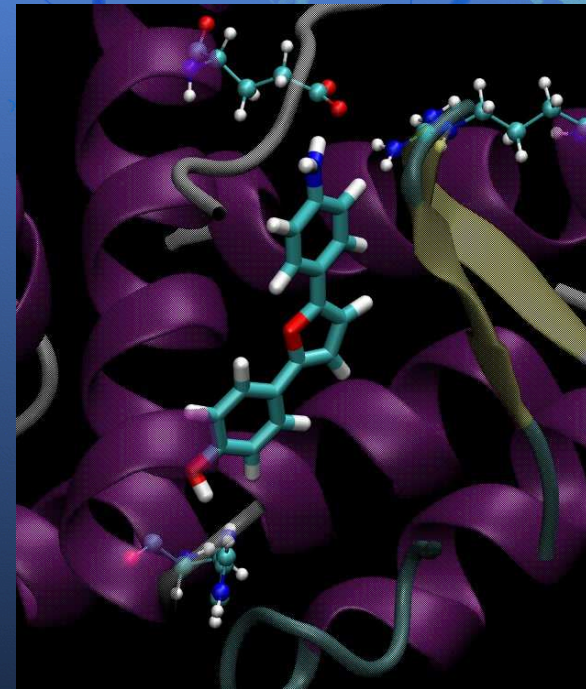


Uncertainties in the binding mode



Binding mode A

“top”



Binding mode B

“bottom”

- Hydrogen donors/acceptors can be satisfied both ways
- Flips observed in some crystal structures



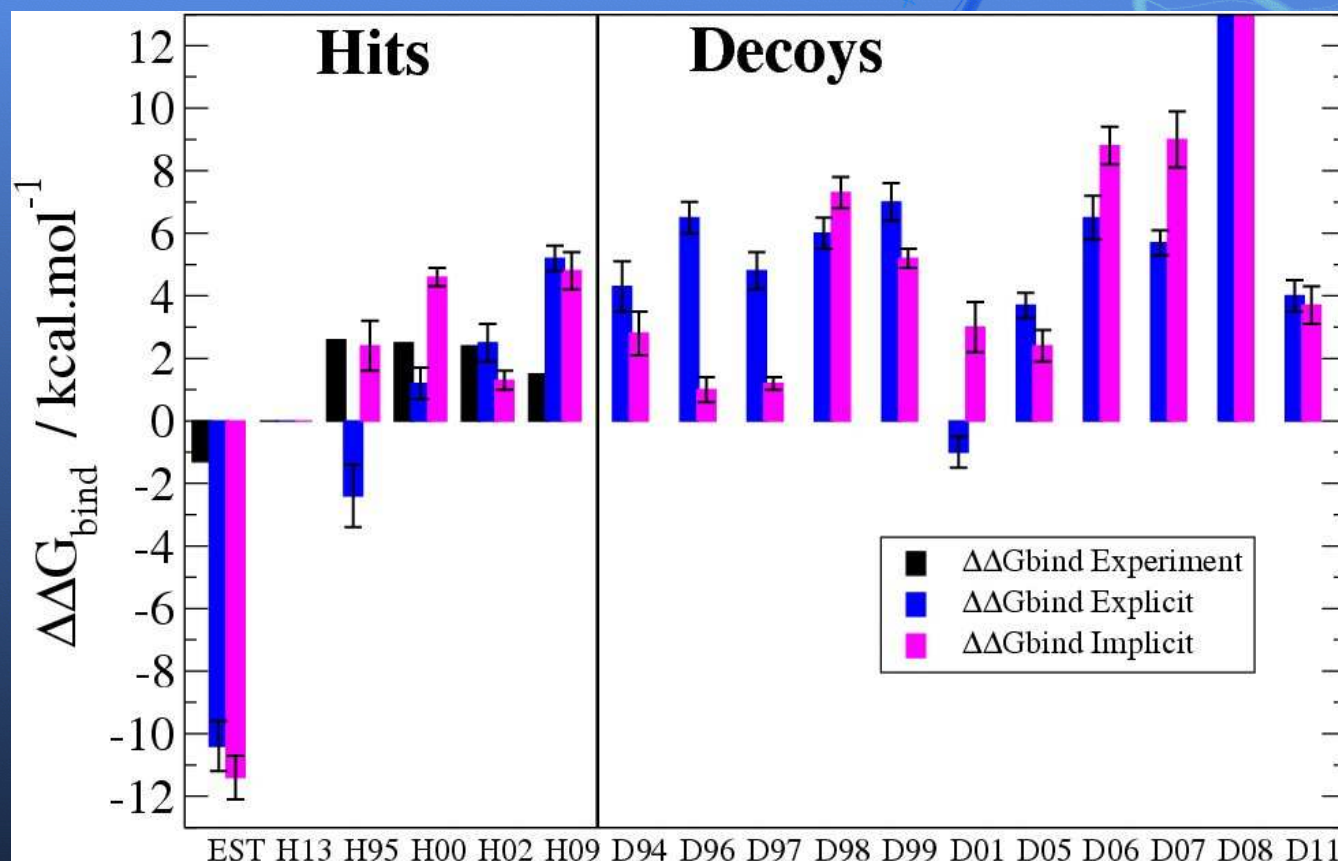
Binding mode predictions

Compound	GOLD	Explicit	Implicit
D94	top	top	top
H95	both	top	top
D96	top	bottom	bottom
D97	top	bottom	both
D98	top	bottom	bottom
D99	top	top	top
H00	top	top	top
D01	top	top	both
H02	top	bottom	bottom
D05	top	top	top
D06	top	bottom	bottom
D07	top	top	top
D08	top	top	top
H09	top	bottom	bottom
D11	top	top	top

- GOLD predicts (almost) always that the hydroxyl group will interact with Glu³⁵³/Arg³⁹⁴
- Alternative orientations are predicted favourable by free energy simulations 6 times
- Implicit solvent simulations suggest both orientations are possible for 2 compounds



Binding affinity predictions



- Explicit solvent: 5 hits in top 6 ligands

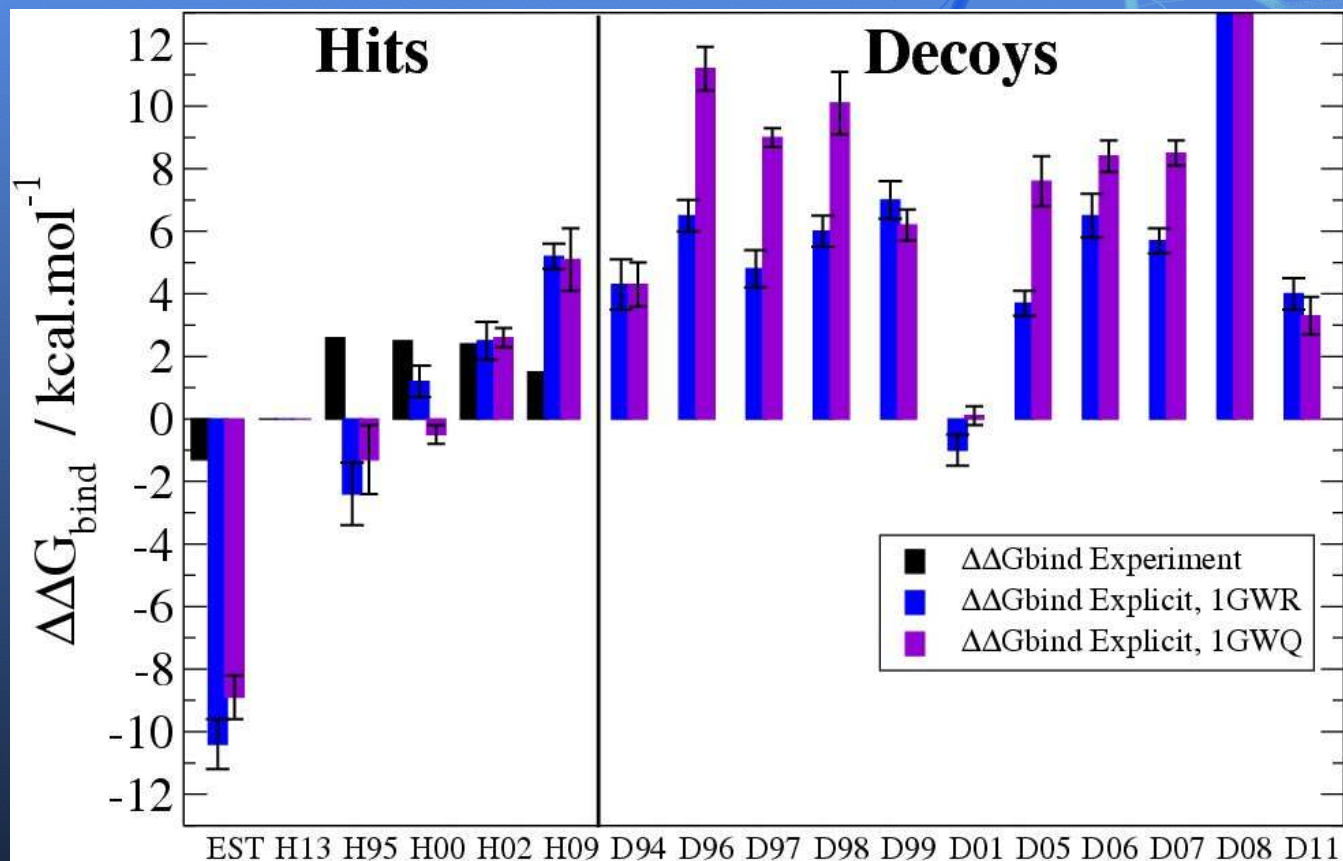
EST < H95 < D01 < H13 < H00 < H02 < D05

- Implicit solvent: 3/4 hits in top 6 ligands

EST < H13 < D96 ~ = H02 ~ = D97 < D05 ~ = H95



Influence of protein structure



- Explicit solvent, 1GWR: 5 hits in top 6 ligands

J. Med. Chem.
in press

EST<**H95**<**D01**<**H13**<**H00**<**H02**<**D05**

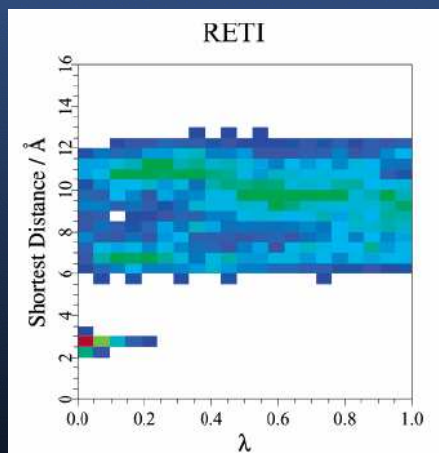
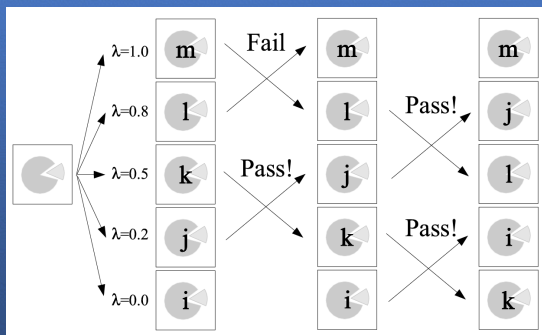
- Explicit solvent, 1GWQ: 5 hits in top 6 ligands

EST<**H95**<**H00**~=**H13**~=**D01**<**H02**<**D11**



Summary

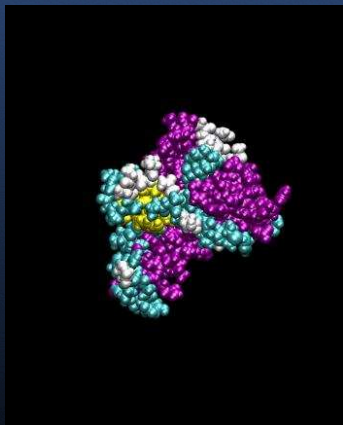
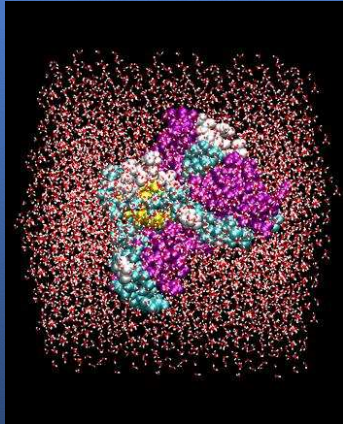
- A binding free energy calculation method
 - Replica exchange





Summary

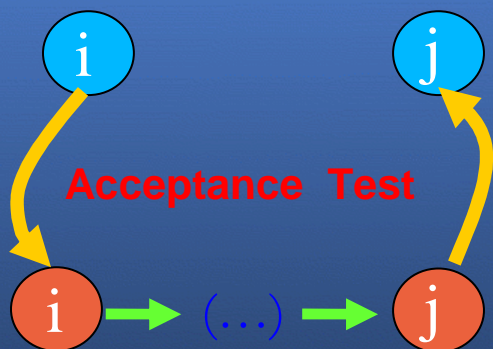
- A binding free energy calculation method
 - Replica exchange
 - **Implicit solvent**





Summary

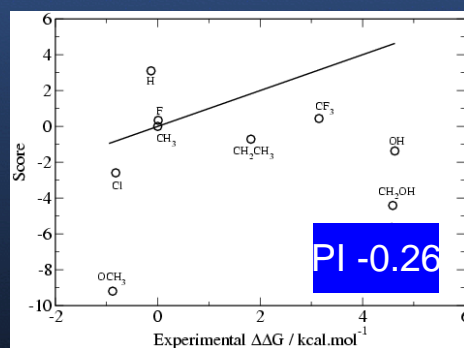
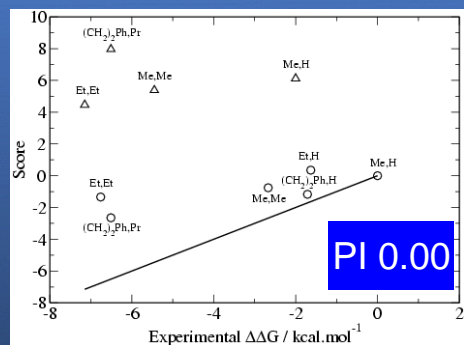
- A binding free energy calculation method
 - Replica exchange
 - Implicit solvent
 - Simplified sampling potential





Summary

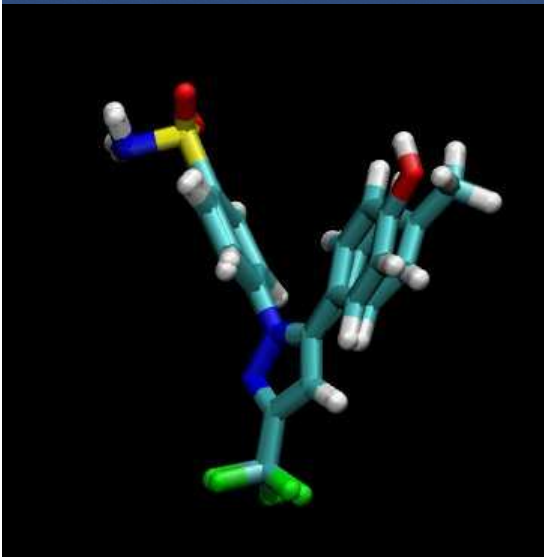
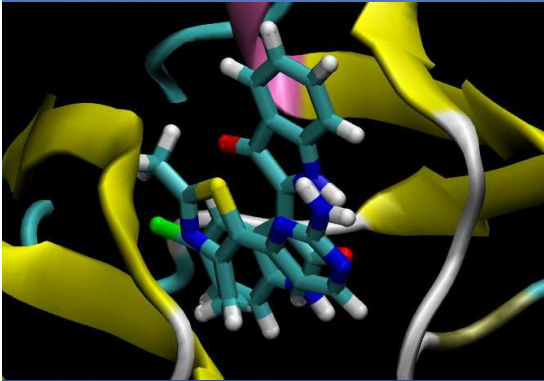
- A binding free energy calculation method
 - Replica exchange
 - Implicit solvent
 - Simplified sampling potential
- Case studies : COX2, Neuraminidase, CDK2, homologous ligands
 - Implicit solvent simulations of comparable or better accuracy than explicit solvent simulations
 - **Better than common scoring functions**





Summary

- A binding free energy calculation method
 - Replica exchange
 - Implicit solvent
 - Simplified sampling potential
- Case studies : COX2, Neuraminidase, CDK2, homologous ligands
 - Implicit solvent simulations of comparable or better accuracy than explicit solvent simulations
 - Better than common scoring functions
 - Problem systems
- **Dual topology – very different molecules**
 - Force field and sampling more critical – cost!





Summary

- A binding free energy calculation method
 - Replica exchange
 - Implicit solvent
 - Simplified sampling potential

Case studies : COX2, Neuraminidase, CDK2, homologous ligands

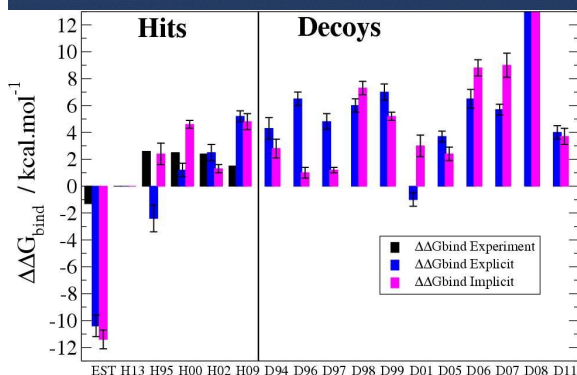
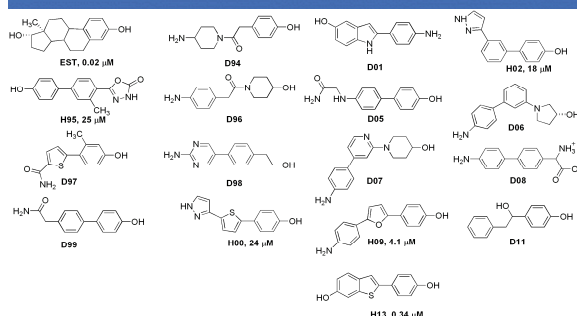
- Implicit solvent simulations of comparable or better accuracy than explicit solvent simulations
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Dual topology – very different molecules

- Force field and sampling more critical – cost!

Broader scope of the methodology

- Binding mode prediction
- Scaffold selection





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- RETI
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- Astex Therapeutics
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